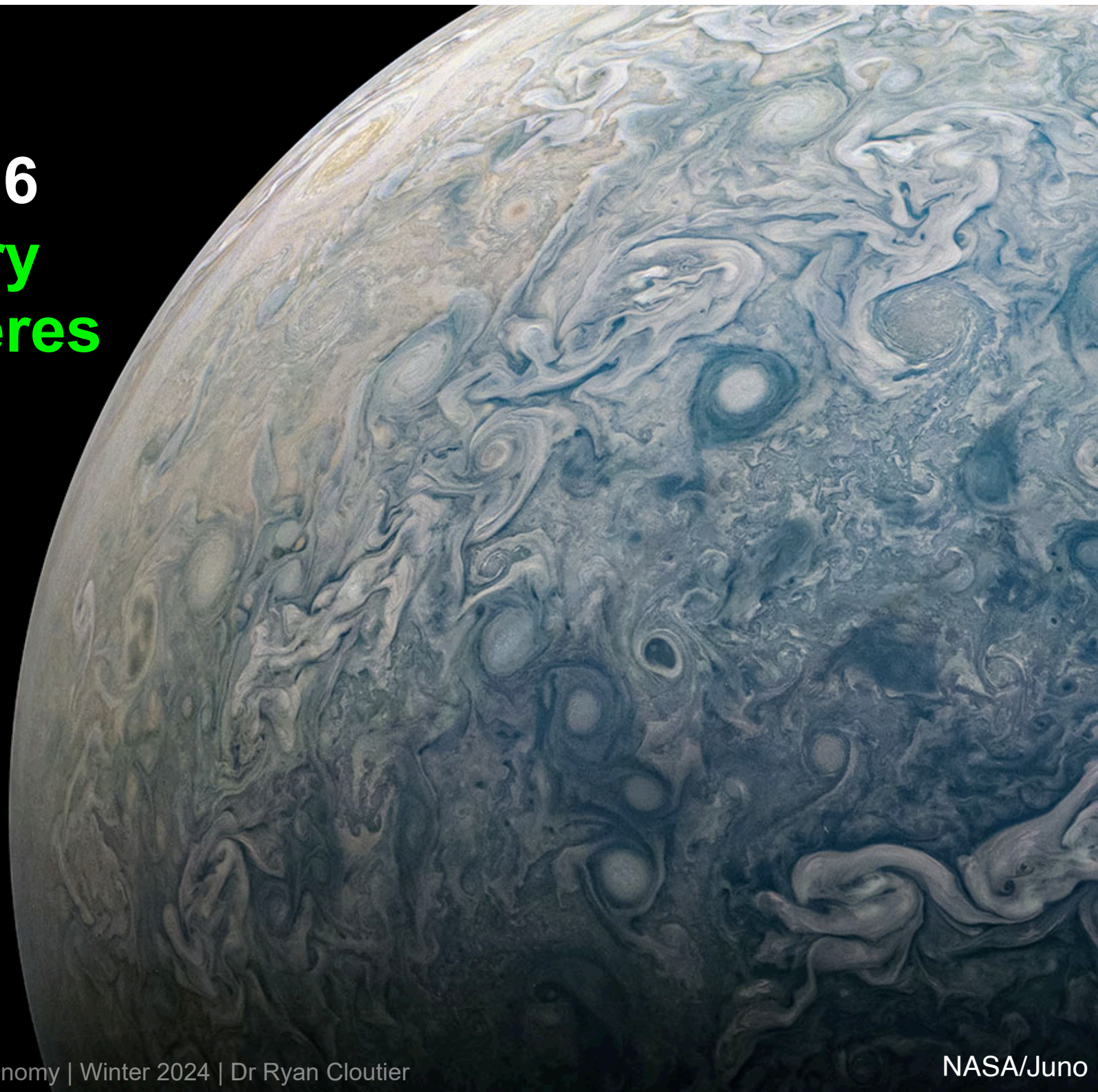


Lecture 6

Planetary Atmospheres



Learning Objectives - Planetary Atmospheres

- 1) Derive the expression for **hydrostatic equilibrium** and use it to derive to **pressure and density profiles** of an Earth-like planet's atmosphere
- 2) Describe the **energy sources** that dictate the **thermal structure** of a planet's atmosphere
- 3) Understand the conditions that lead to **atmospheric convection** and evaluate where atmospheres are convective based on their thermal structures
- 4) List the **main chemical constituents** of the planetary atmospheres in the solar system and on exoplanets from **transmission spectroscopy**
- 5) Outline the physics behind the **cloud formation** process
- 6) Outline the physics behind **winds** in the solar system and beyond

Atmospheres

The **gaseous outer layers** of planets and moons

Terrestrial planet atmospheres

- thin
- mass fractions of $\lesssim 10^{-3}$
- dominated by heavy species **N₂, CO₂, and others**

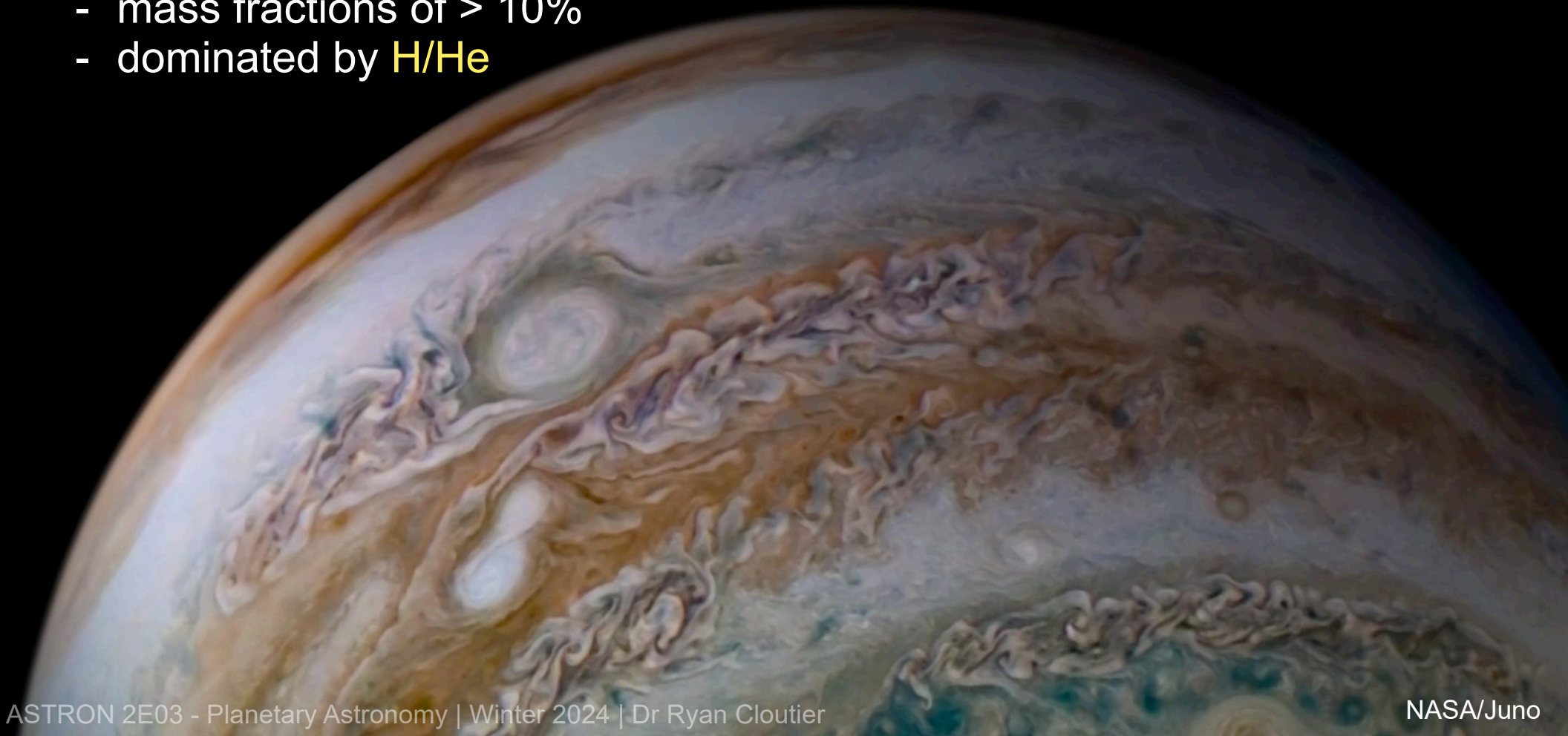


Atmospheres

The **gaseous outer layers** of planets and moons

Giant planet atmospheres

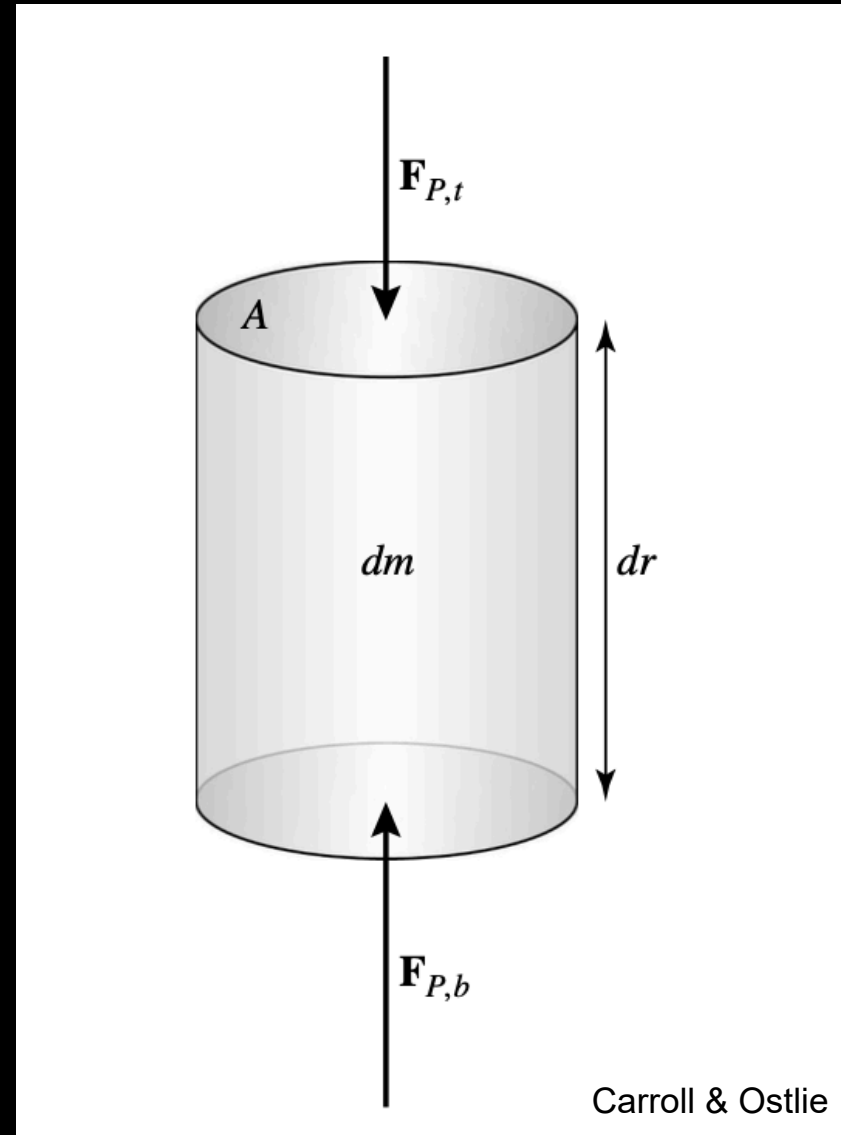
- deep
- mass fractions of $> 10\%$
- dominated by **H/He**



Hydrostatic Equilibrium (HSE)

Atmospheres are held up against the force of gravity F_g by a gas pressure gradient

$$\Delta P = \frac{F_{P,t} - F_{P,b}}{\Delta A}$$



Hydrostatic Equilibrium (HSE)

In class we'll derive the **expression for HSE** given below

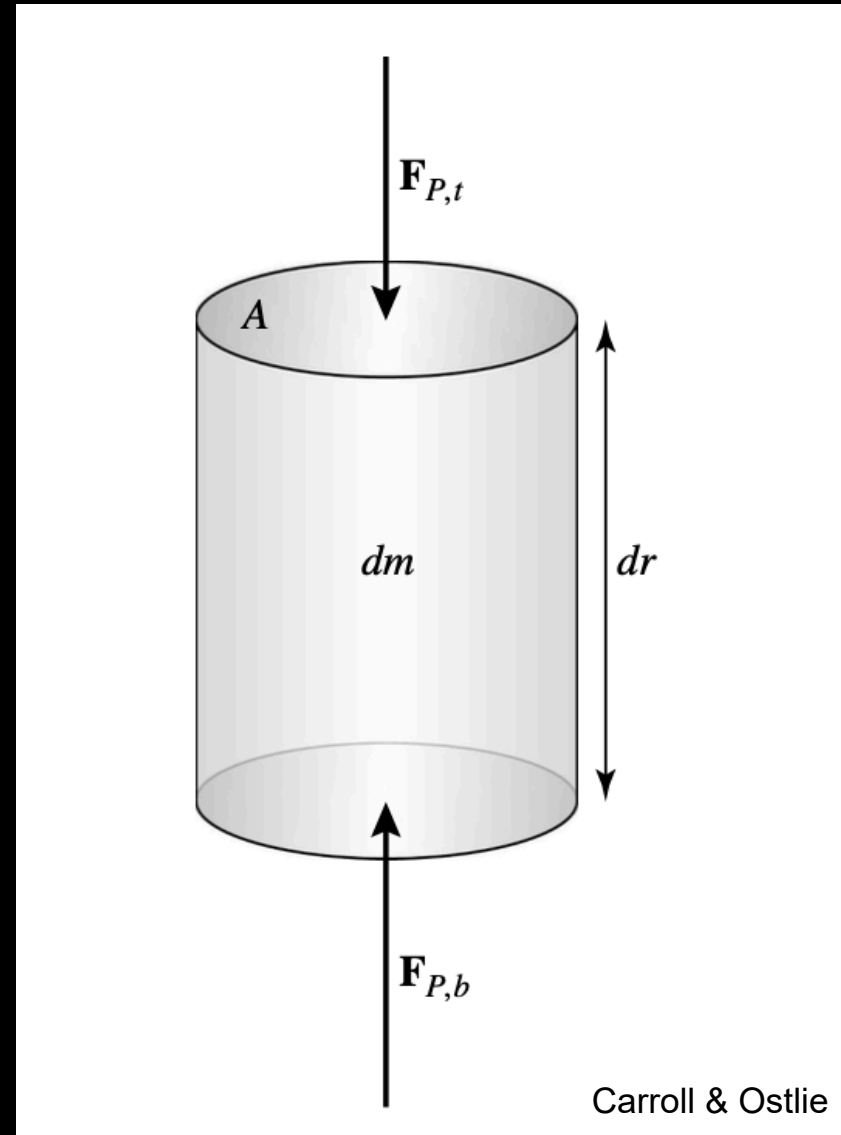
$$\frac{dP}{dr} = -\rho g$$

P : fluid pressure

r : vertical coordinate (i.e. height of the atmosphere)

ρ : fluid density

g : gravitational acceleration



Atmospheric structure

We're interested in knowing how the **atmospheric pressure changes with height** in the atmosphere.

To do this, we need to write down an **equation of state (EOS)** for the atmospheric gas

$$P(\rho, T)$$

An EOS typically has the form above.

It describes the relationship between a material's **pressure, density, and temperature**

Atmospheric structure

In low to moderate pressure environments, **typical of terrestrial planet atmospheres**, the atmosphere is well-described as an **ideal gas**

$$P = \frac{\rho k_B T}{\mu m_H}$$

P : fluid pressure

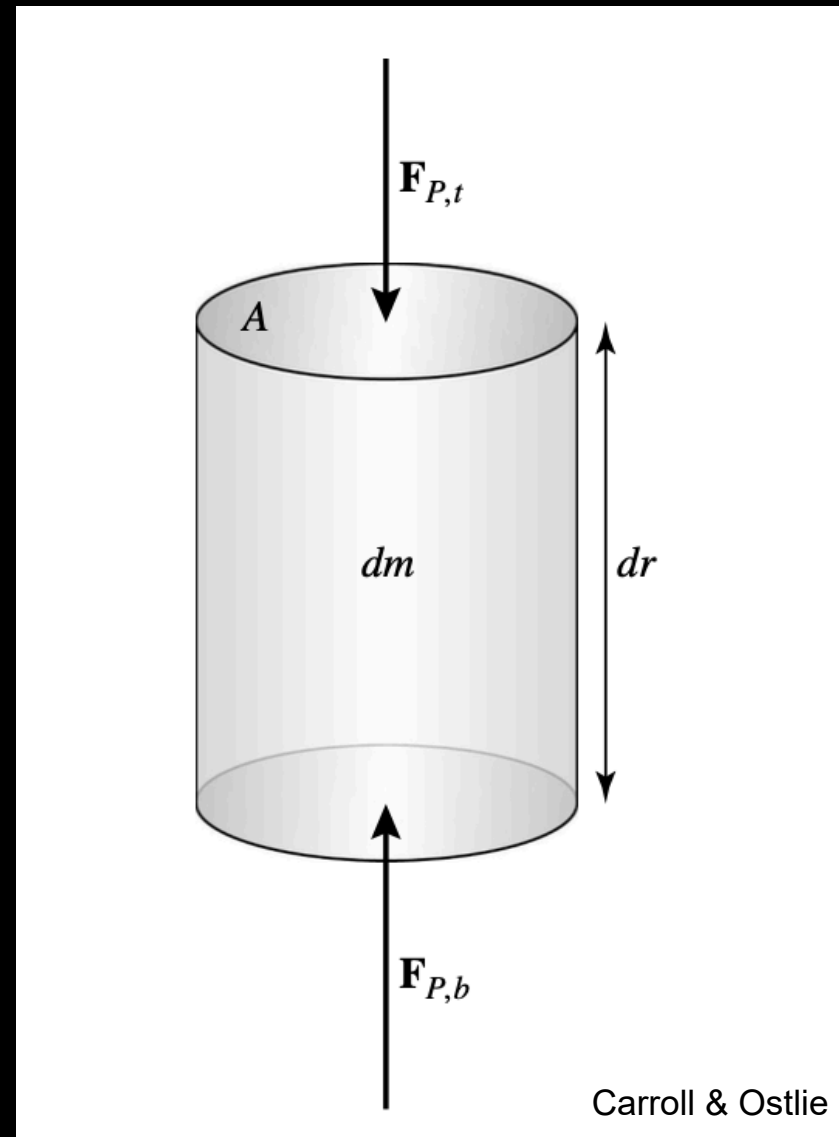
k_B : Boltzmann constant

T : fluid temperature

ρ : fluid density

μ : mean molecular weight

m_H : mass of the hydrogen atom



Mean Molecular Weight (μ)

Atomic mass unit ~ the proton mass ~ the neutron mass = $1.67e-27$ kg

Gas	Composition	Mean molecular weight μ
Molecular nitrogen	$N_2 = 2*N$	$2*14 = 28$
Molecular oxygen	$O_2 = 2*O$	$2*16 = 32$
Water	$H_2O = 2*H + O$	$2*1 + 16 = 18$
Earth atmosphere	78% N_2 + 22% O_2	$0.78*28 + 0.22*32 \sim 29$

Atmospheric Structure

In class we'll show that for an **atmosphere in HSE** and composed of an **ideal gas**, the atmospheric **pressure** and **density profiles** are

$$P(r) = P_0 e^{-r/H}$$

$$\rho(r) = \rho_0 e^{-r/H}$$

where the subscript 0 indicates the values at the **surface** and H is the atmospheric **scale height**

$$H = \frac{k_B T}{\mu m_H g}$$

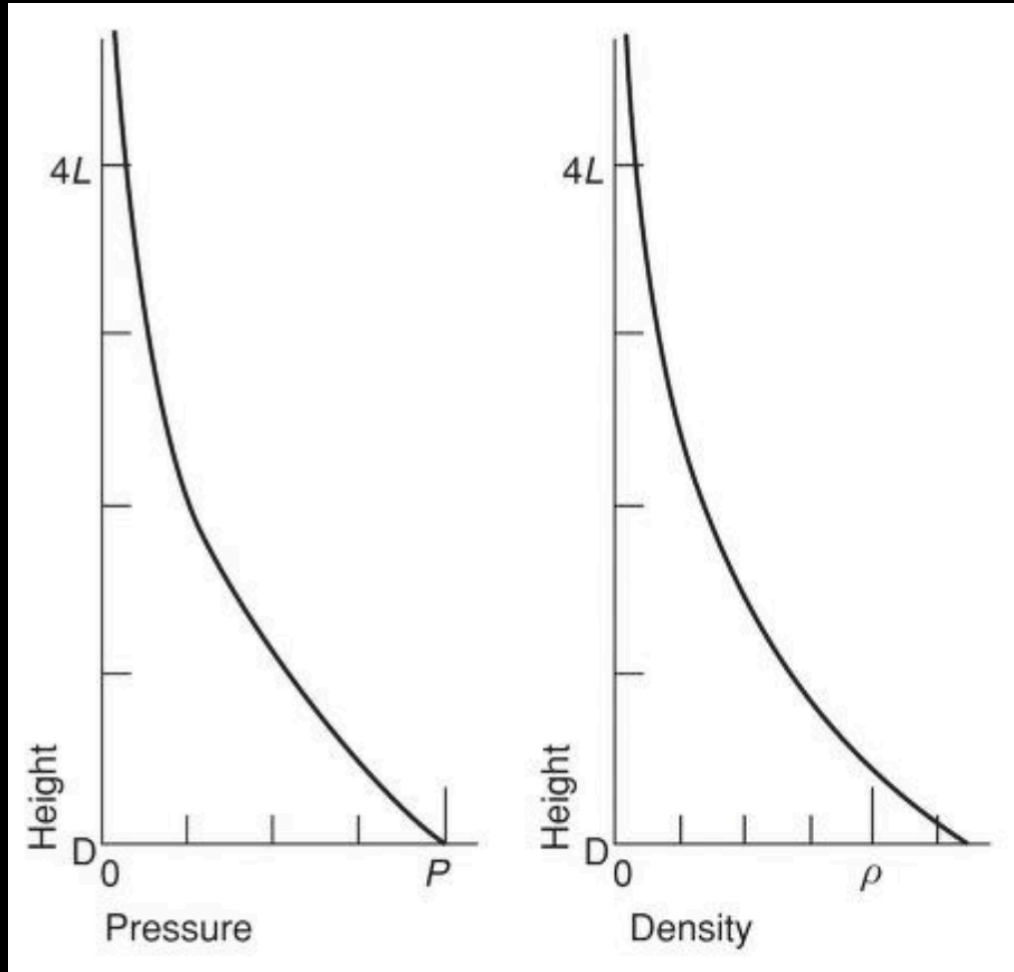
Atmospheric Structure

$$P(r) = P_0 e^{-r/H}$$

$$\rho(r) = \rho_0 e^{-r/H}$$

where

$$H = 8.3 \text{ km} \left(\frac{T}{288 \text{ K}} \right) \left(\frac{g}{9.8 \text{ m/s}^2} \right)^{-1} \left(\frac{\mu}{29 m_H} \right)^{-1}$$



Carroll & Ostlie

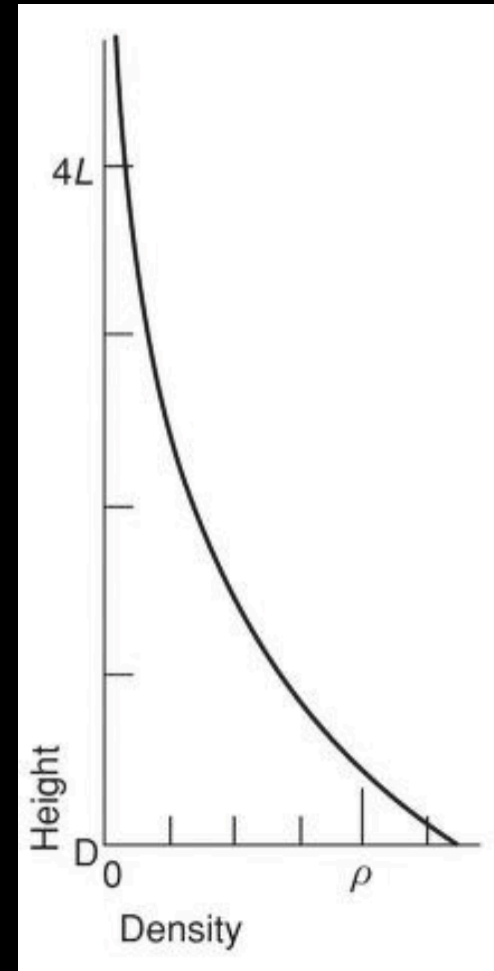
Atmospheric Mass

Note that most of the atmosphere's mass is within **one scale height H**

$$\rho(r) = \rho_0 \exp(-r/H)$$

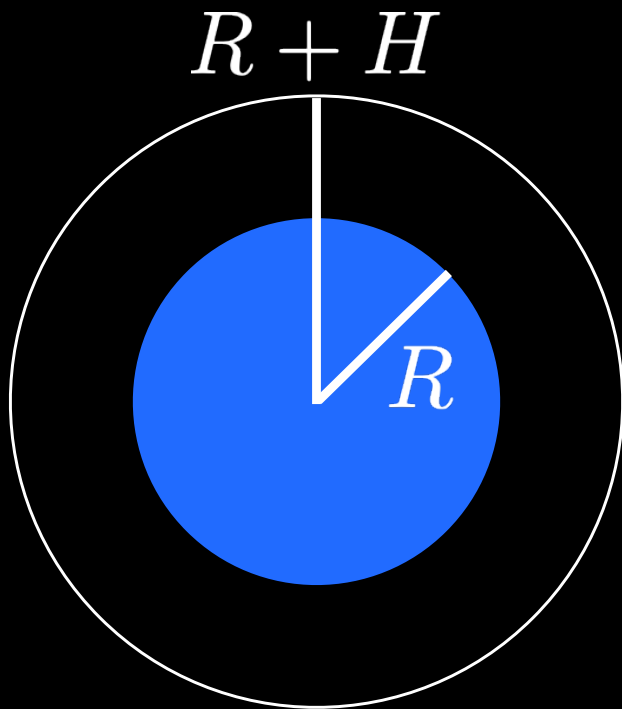
where

$$H = 8.3 \text{ km} \left(\frac{T}{288 \text{ K}} \right) \left(\frac{g}{9.8 \text{ m/s}^2} \right)^{-1} \left(\frac{\mu}{29 m_H} \right)^{-1}$$



Carroll & Ostlie

Atmospheric Mass



Normally, one would calculate the atmospheric mass by **integrating the atmospheric density profile**

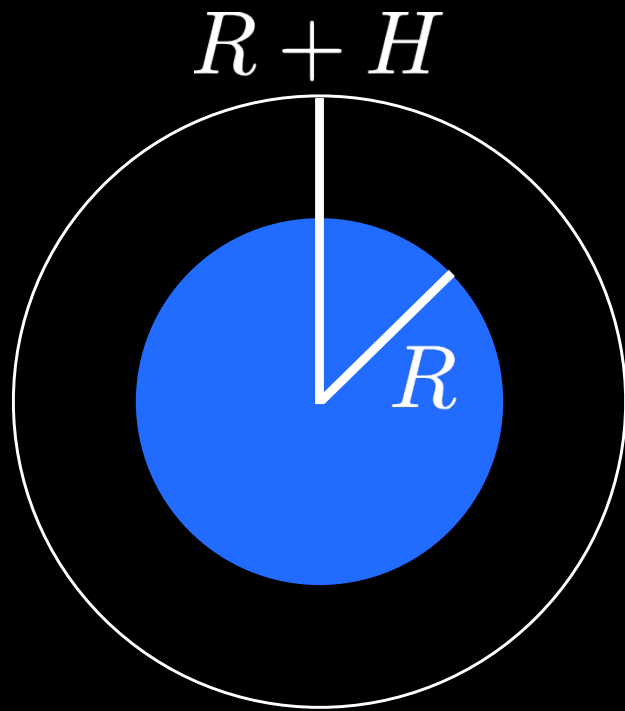
$$\rho(r) = \rho_0 e^{-r/H}$$

from the planet's surface to H .
(noting that that $\rho_0 = 1.2 \text{ kg/m}^3$)

Atmospheric Mass

But we can approximate the atmospheric mass by noting that most of the mass is contained within $r < H$, with $\rho_0 = 1.2 \text{ kg/m}^3$.

The atmospheric mass is then

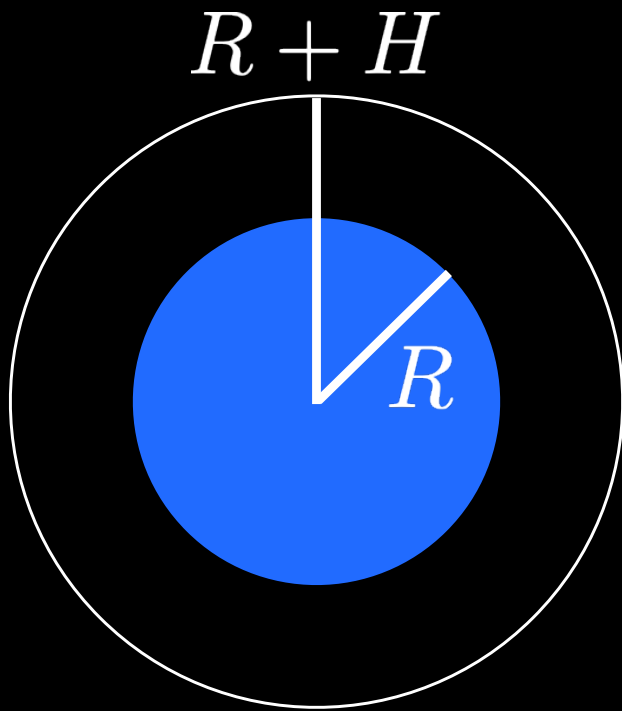


$$\int_0^{M_{atm}} dM \approx \rho_0 dV$$

$$\begin{aligned} M_{atm} &\approx \rho_0 \int_R^{R+H} 4\pi r^2 dr \\ &= \frac{4\pi}{3} \rho_0 [(R+H)^3 - R^3] \\ &= 5 \times 10^{18} \text{ kg} \end{aligned}$$

Actual is $5.1 \times 10^{18} \text{ kg!!}$

Atmospheric Mass



On terrestrial planets for which $H \ll R$, the atmospheric mass can be approximated as

$$M_{atm} = M_{atm,\oplus} \left(\frac{\rho}{\rho_{\oplus}} \right) \left(\frac{H}{H_{\oplus}} \right)$$

Thermal Structure

Note that in deriving the **pressure** and **density profiles**

$$P(r) = P_0 e^{-r/H}$$

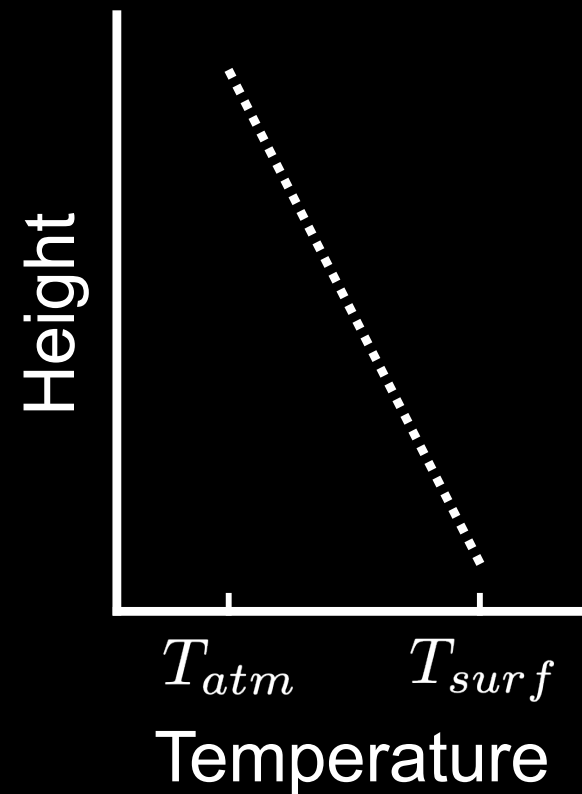
$$\rho(r) = \rho_0 e^{-r/H}$$

we implicitly assumed that the atmospheric temperature profile was **isothermal**

Thermal Structure

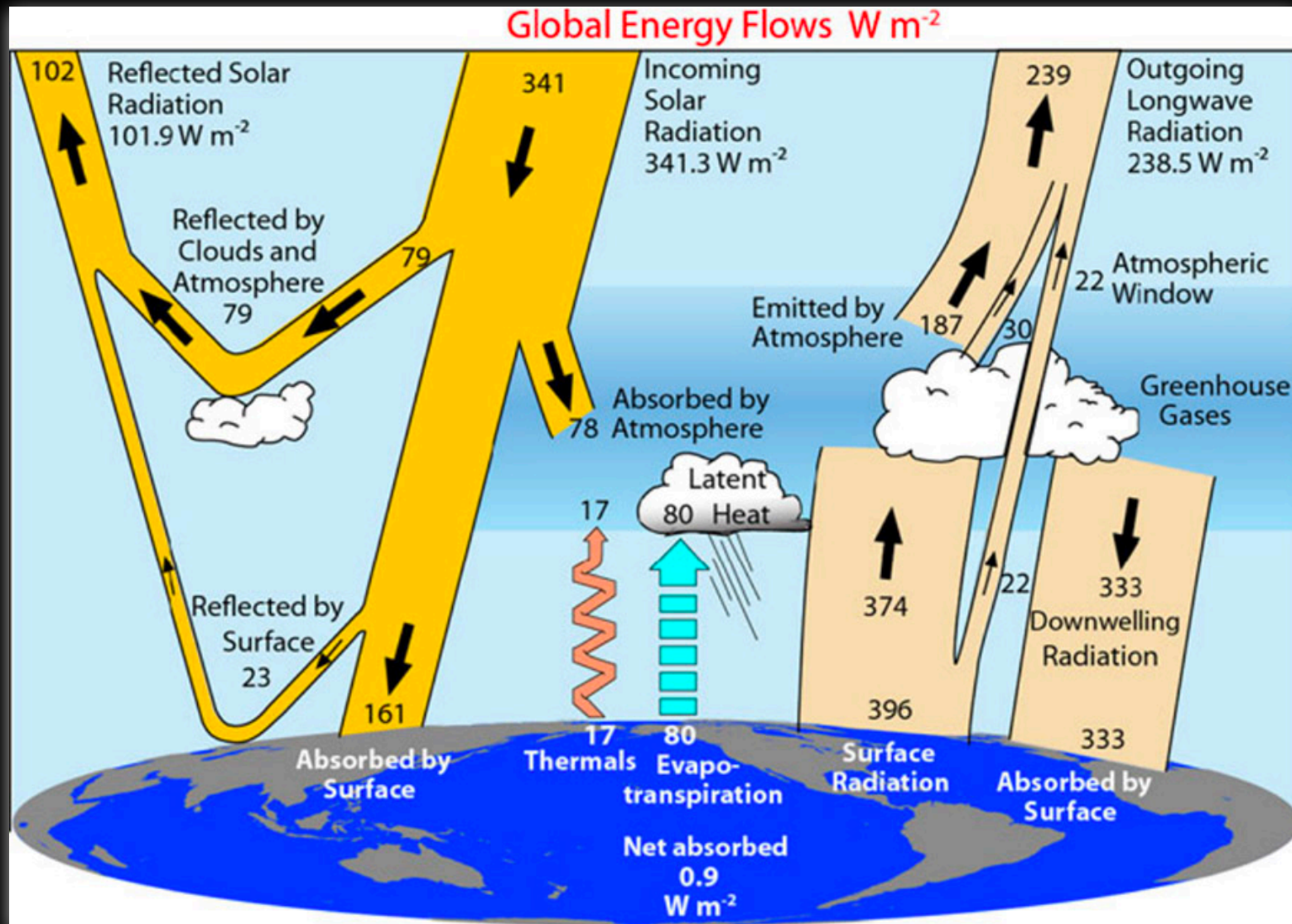
But recall from our discussion of that **greenhouse effect** that

$$\frac{dT}{dr} < 0$$



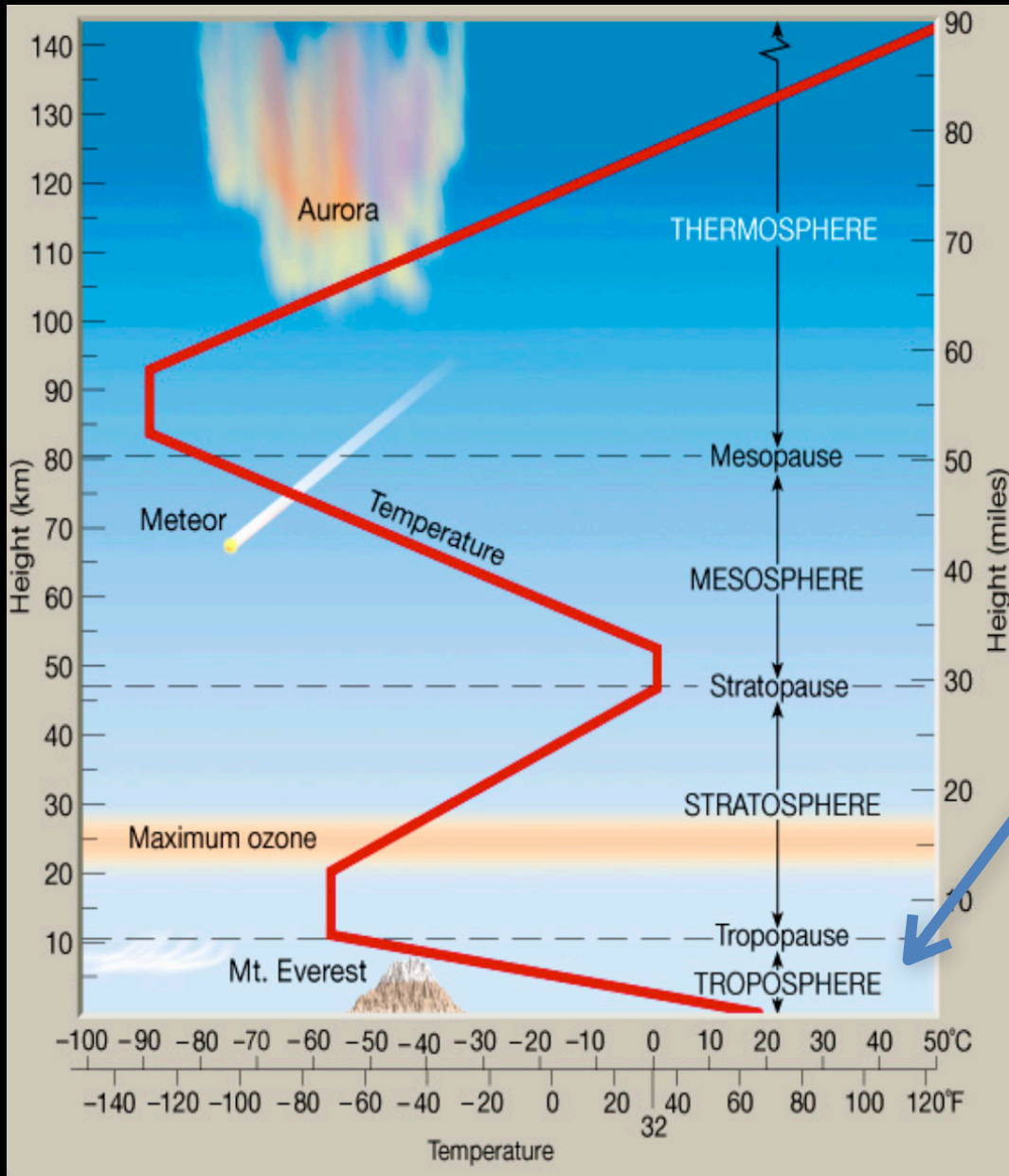
We must consider **all available energy sources**

Radiative feedback alters the atmospheric temperature profile



Kevin Trenberth, John Fasullo and Jeff Kiehl

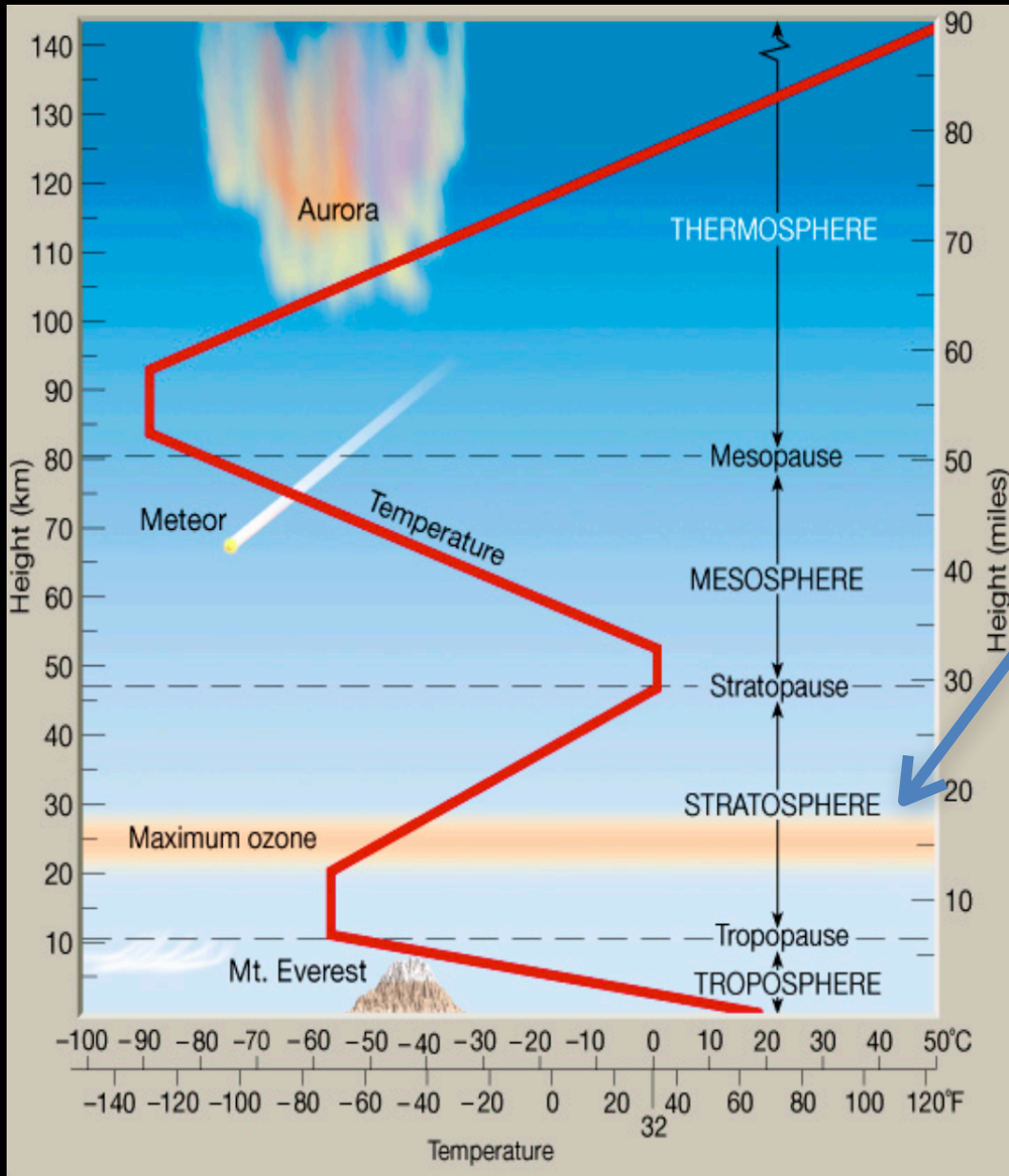
Earth's Atmosphere



TROPOSPHERE

- **Densest portion** containing most of atmospheric mass below $H = 8.3$ km
- Rich in **GH gases** (H_2O , CO_2 , CH_4)
- Energy transport is via **convection**

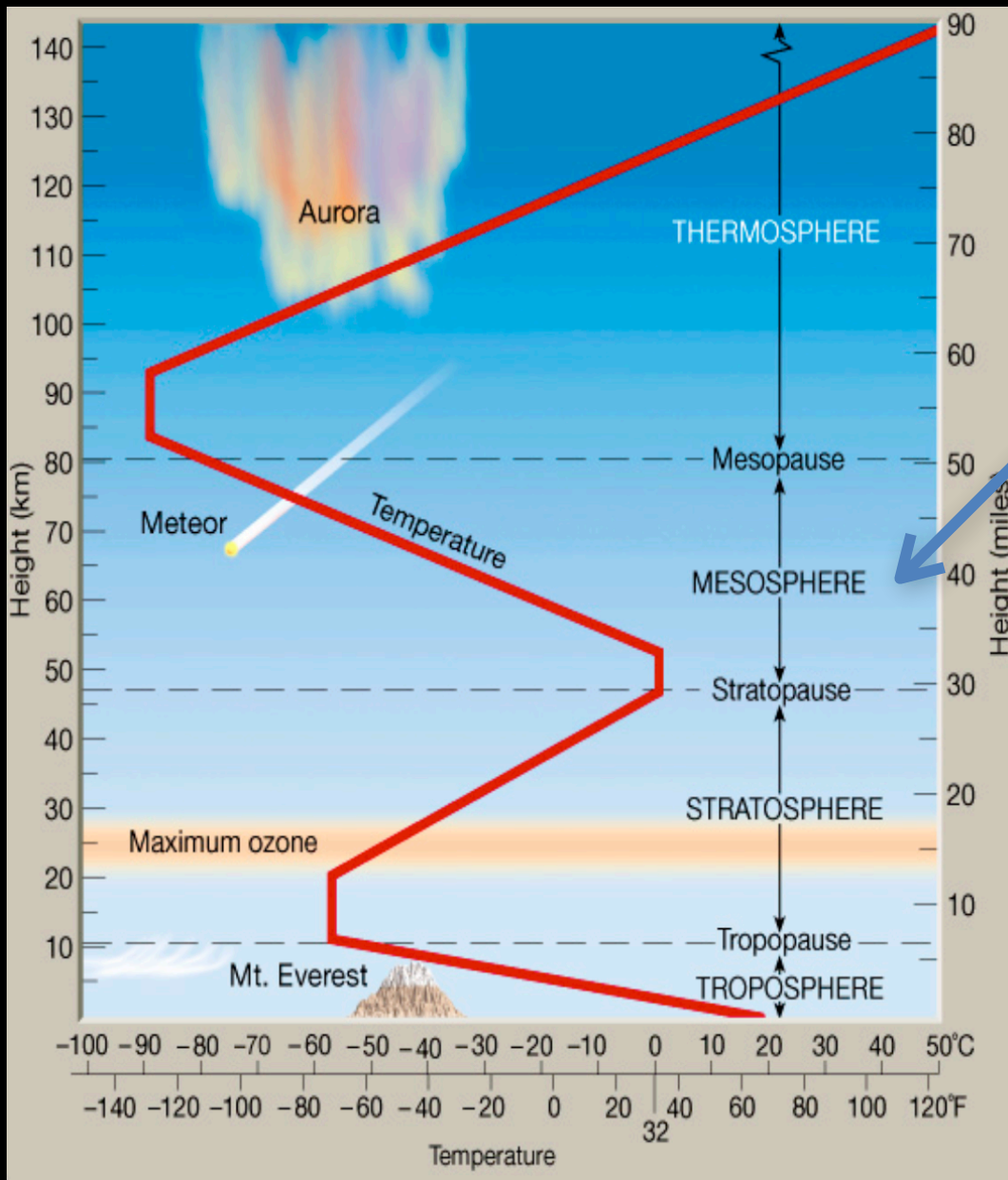
Earth's Atmosphere



STRATOSPHERE

- **Thermal inversion** produced by UV absorption of O_3

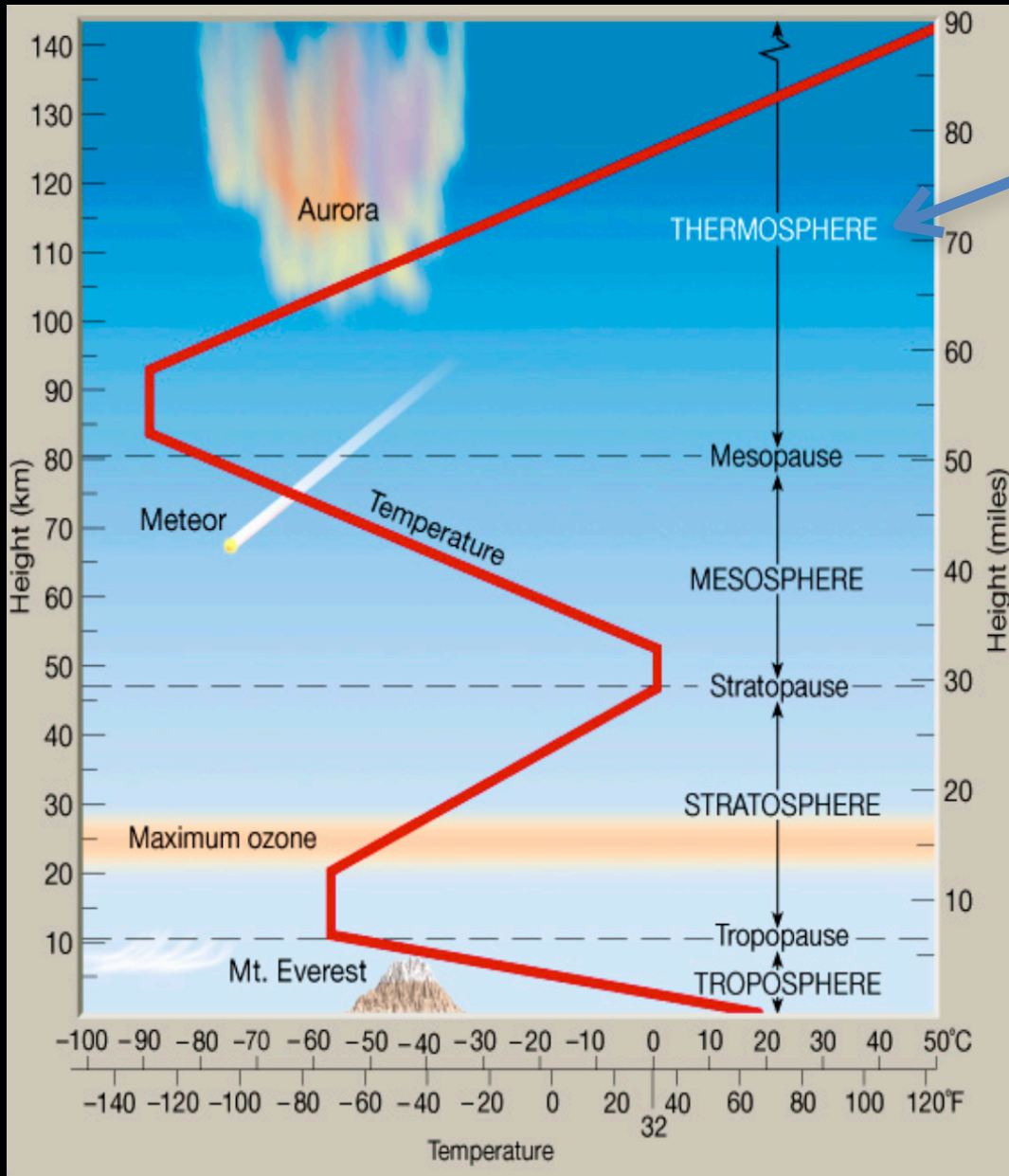
Earth's Atmosphere



MESOSPHERE

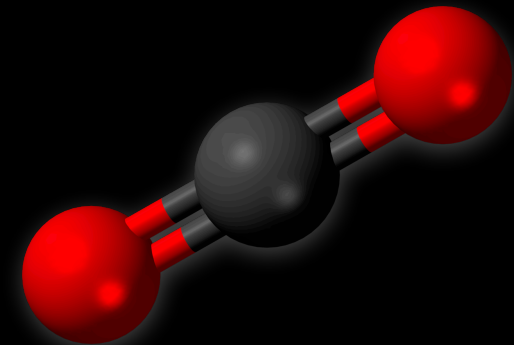
- Lower O₃ abundance results in **inefficient UV heating**
- GH gases do not “see” as much longwave radiation as in the troposphere, such that the **GH effect is weak**

Earth's Atmosphere



THERMOSPHERE

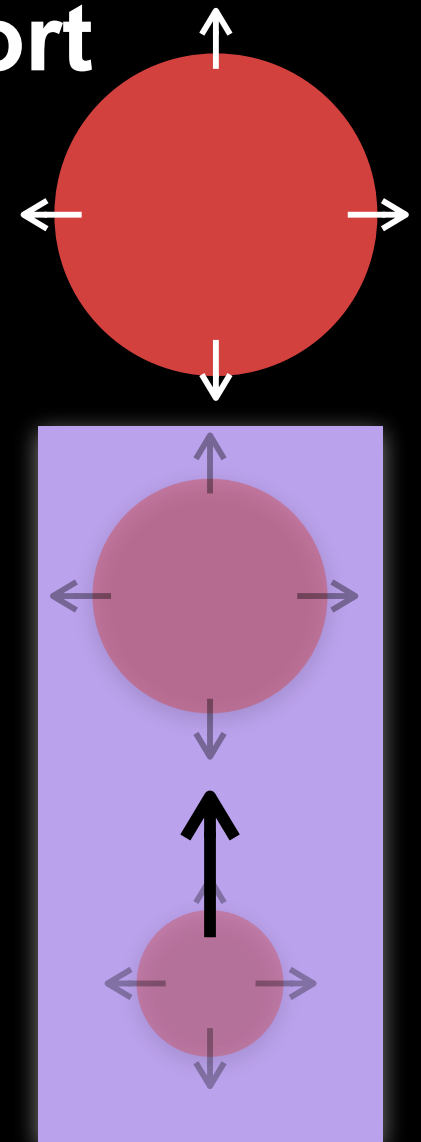
- **UV heating** by O_2
- **Auroral heating** by the interactions with the solar winds
- Atmosphere density is low such that molecular **cooling is inefficient**



Convective energy transport

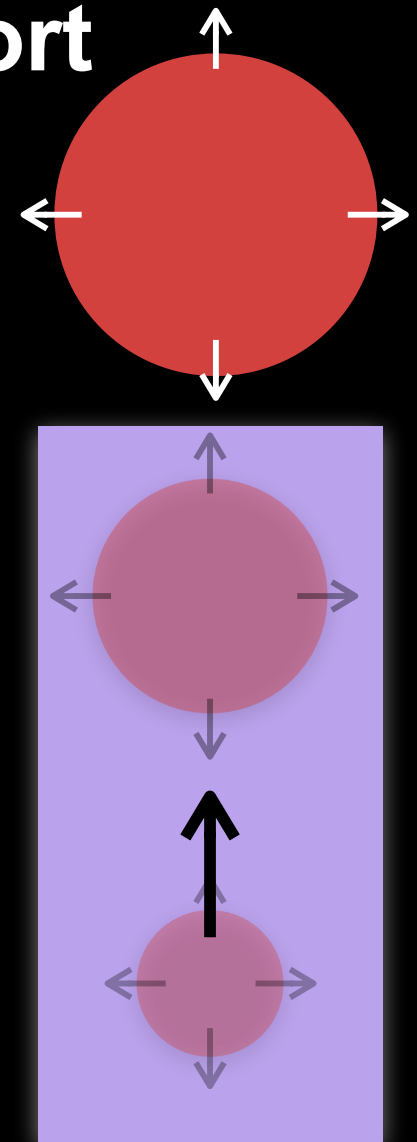
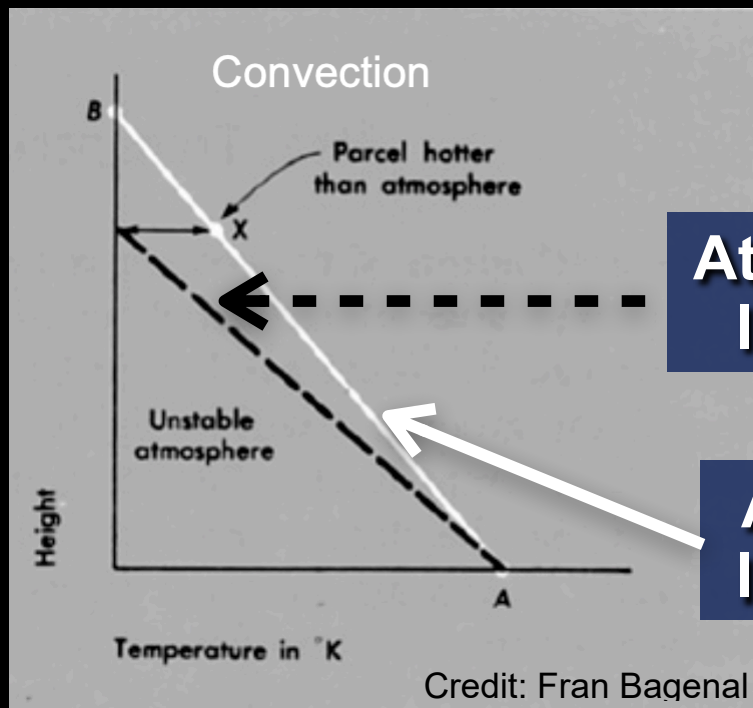
Consider a **parcel of air** that is hotter than its surroundings

- according to the ideal gas law, $P_{parcel} > P_{surroundings}$ so the **parcel expands** to reach a pressure balance
- its lower density causes the **parcel to rise** (buoyancy)
- because P decreases with altitude, the rising parcel continues to **expand and cool as it rises**
- if cooling is slow, then the parcel will **remain warmer than its surroundings** and continue to **rise**, carrying thermal energy with it (i.e. convective heat transport)



Convective energy transport

An atmosphere is said to be **unstable against convection** (i.e. it is convective) if its temperature drops rapidly with altitude compared to the change in parcel temperature



Adiabatic lapse rate

The cooling rate of a rising parcel of gas

$$\left(\frac{dT}{dr}\right)_{\text{ad}} = -\frac{\gamma - 1}{\gamma} \left(\frac{g\mu m_H}{k_B}\right)$$

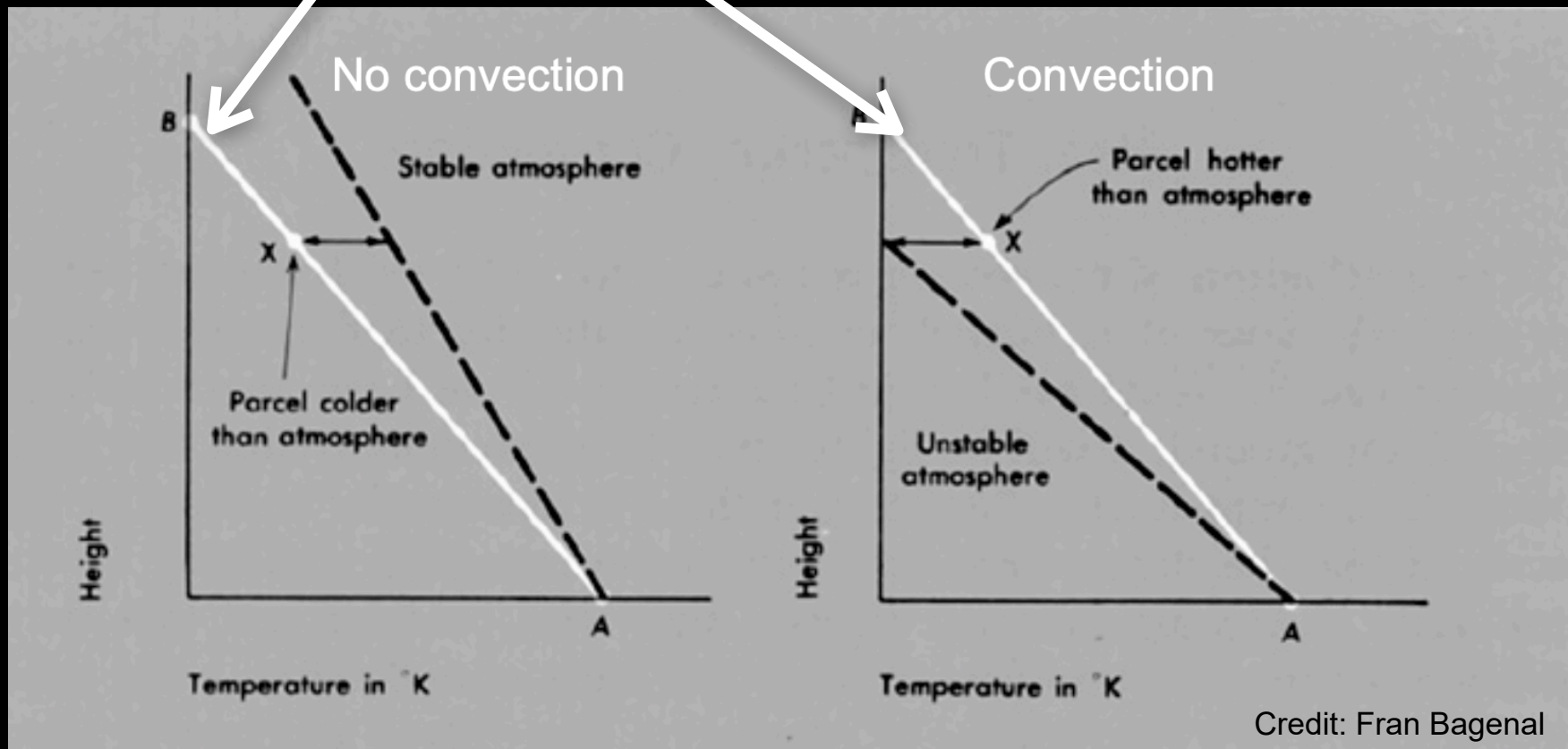
Depends on the planet's **surface gravity**, **mean molecular weight**, and the **adiabatic index γ** describing the heat required to raise the gas temperature at a fixed P and V .

$\gamma = 5/3, 7/5, 4/3$ for monoatomic, diatomic, and polyatomic gases respectively

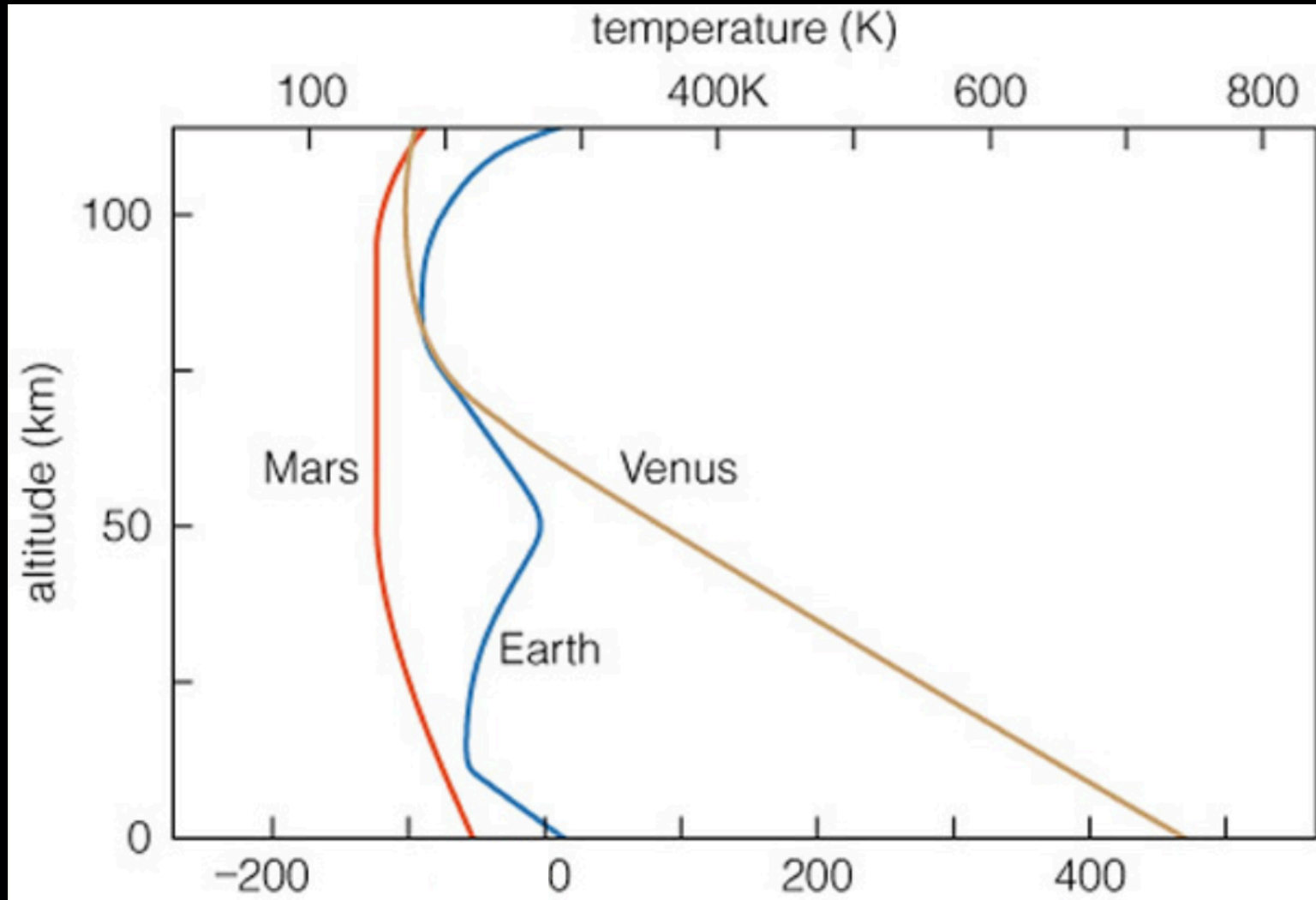
Adiabatic lapse rate

The cooling rate of a rising parcel of gas

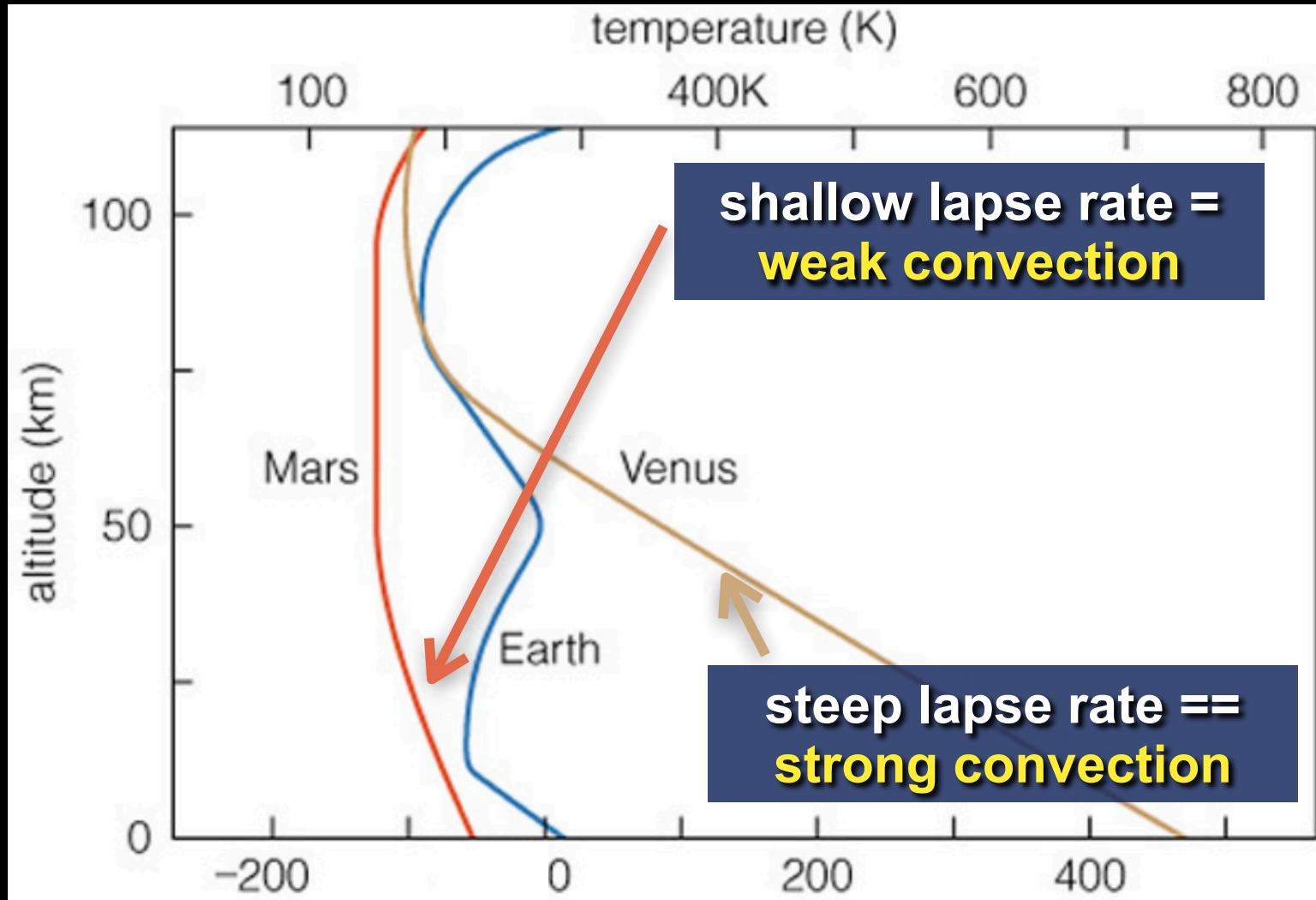
$$\left(\frac{dT}{dr}\right)_{\text{ad}} = -\frac{\gamma - 1}{\gamma} \left(\frac{g\mu m_H}{k_B}\right)$$



Comparative Atmospheres



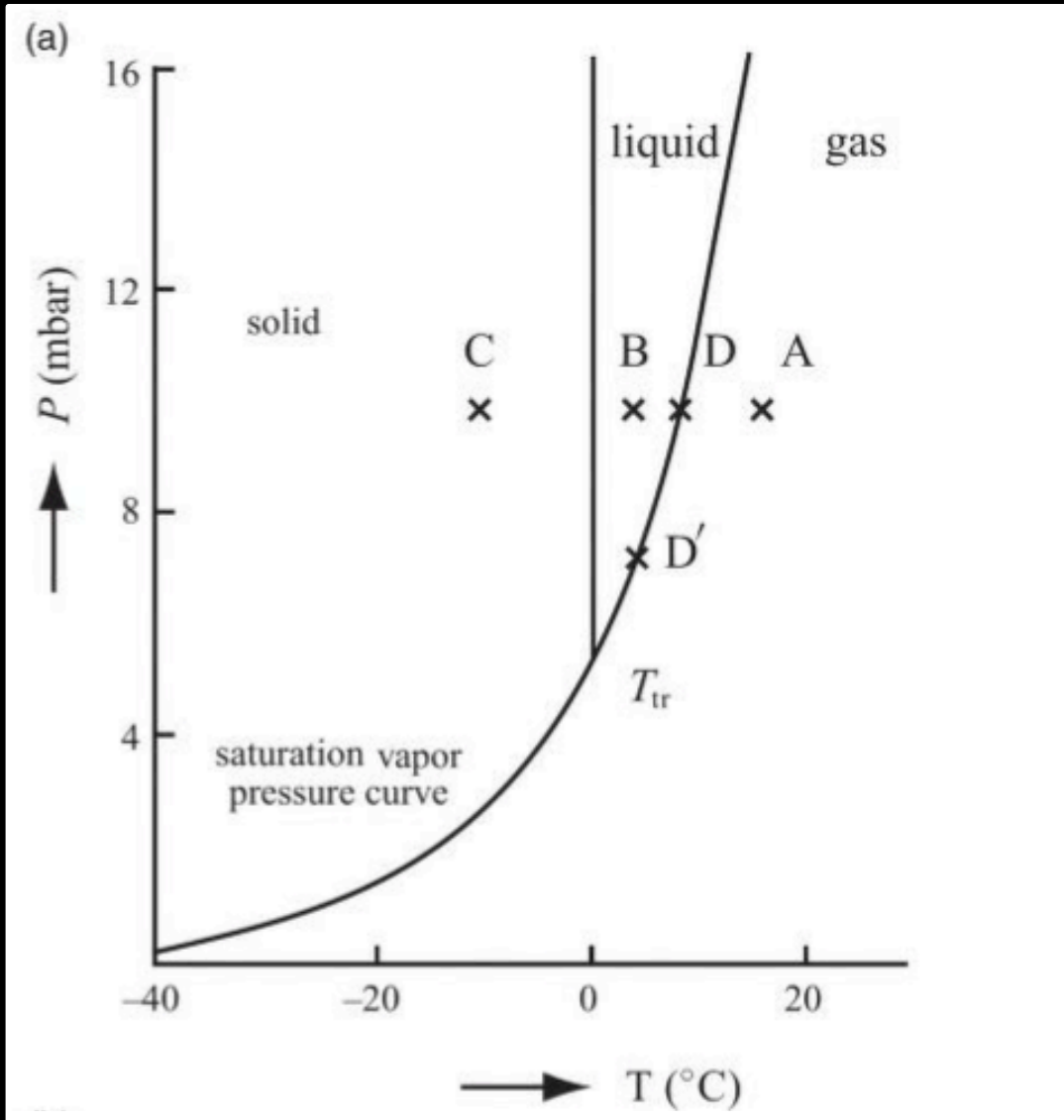
Comparative Atmospheres



Cloud formation modifies the atmospheric lapse rate



Cloud Formation



Lissauer & de Pater

At a fixed P and T , air cannot contain more water than the **saturation vapour pressure**, which is given by the **Clausius-Clapeyron EOS**

$$P = C_L e^{-L/(R_{gas}T)}$$

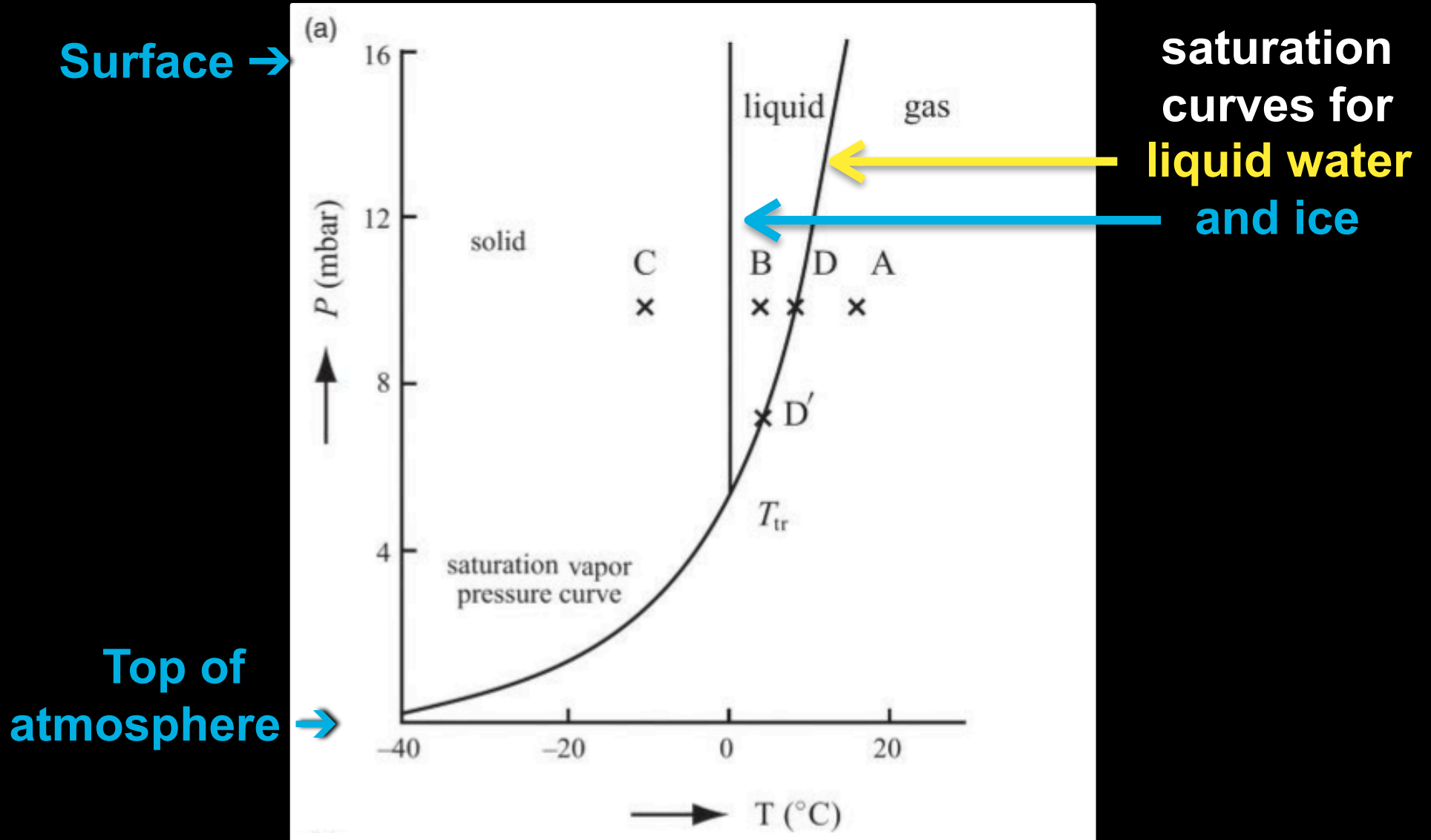
P : saturation vapour pressure of water

C_L, R_{gas} : gas constants

L latent heat

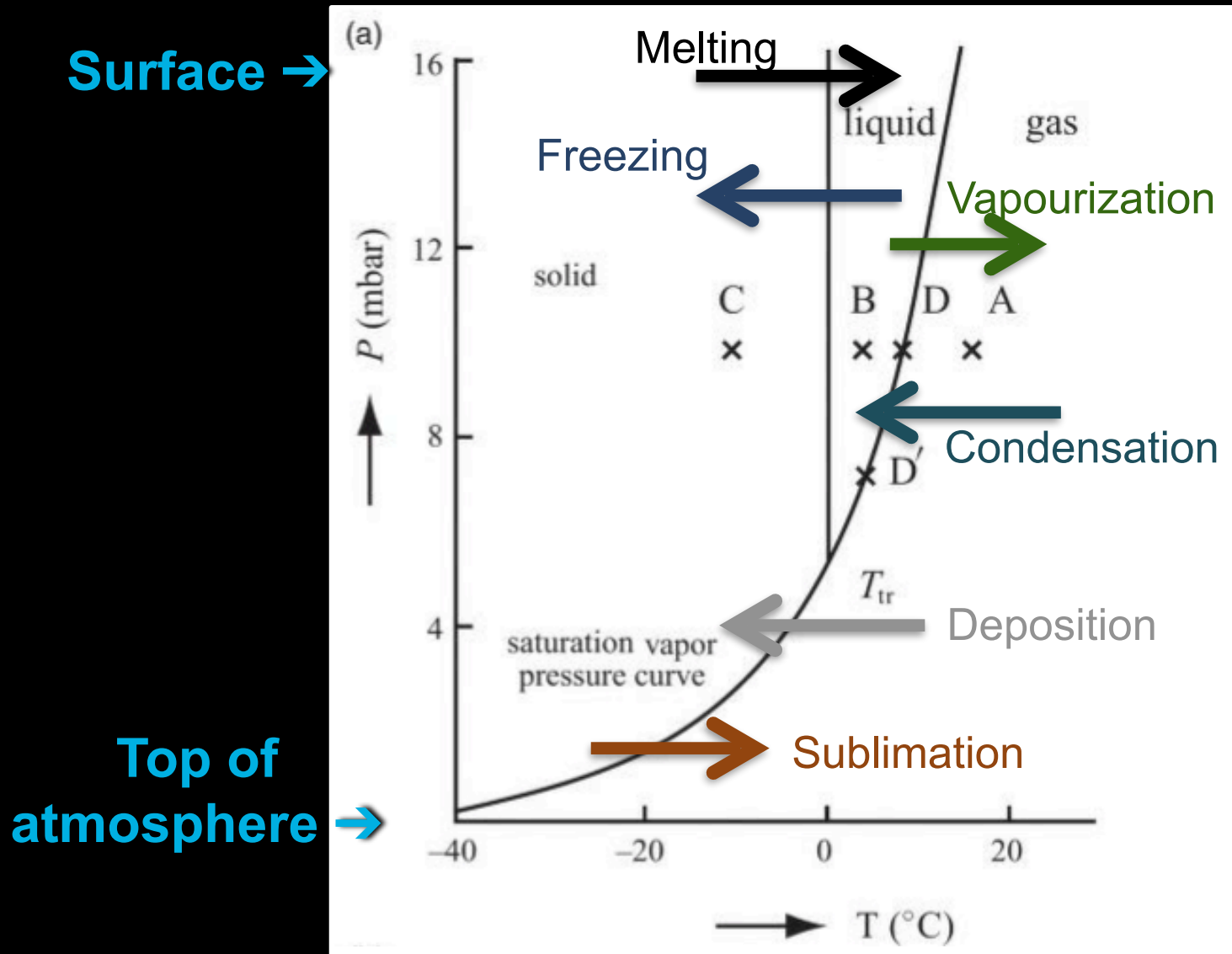
T : air temperature

Cloud Formation



Lissauer & de Pater

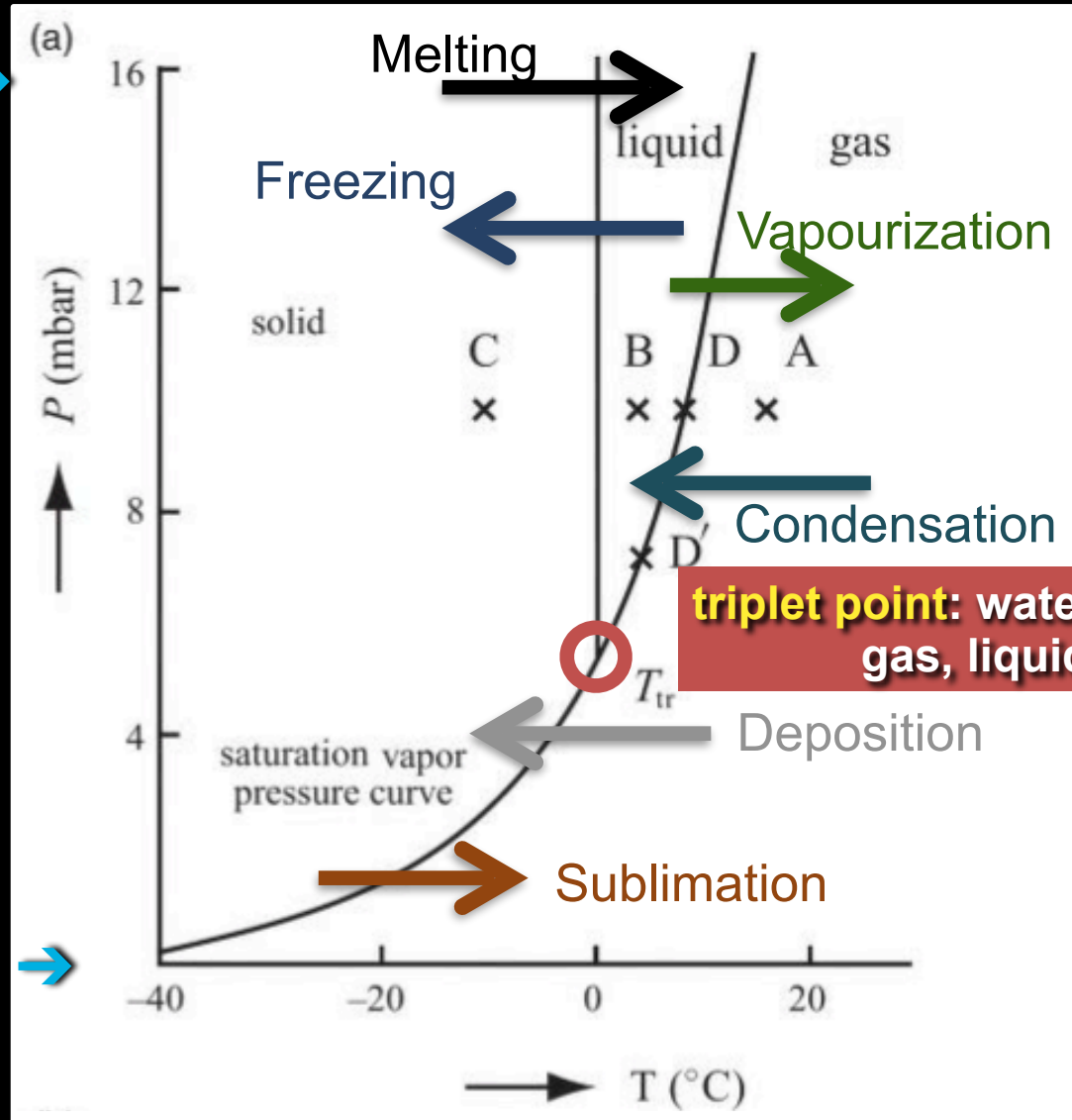
Cloud Formation



Lissauer & de Pater

Cloud Formation

Surface →



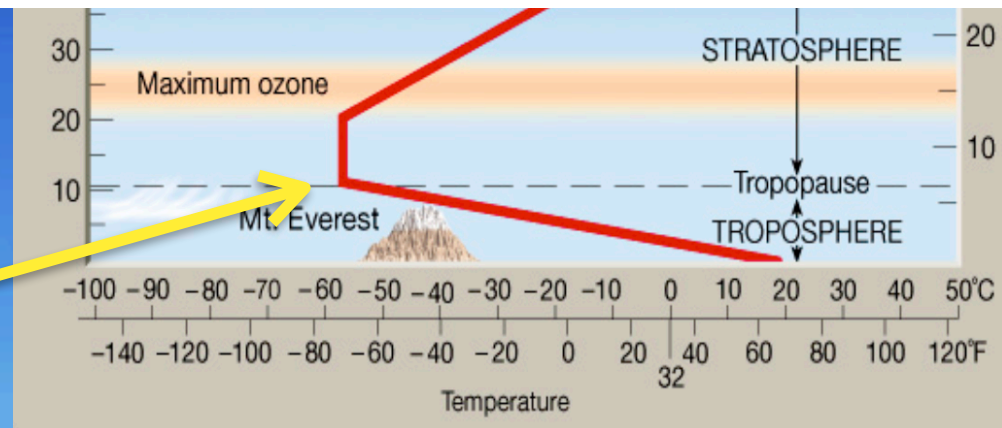
triplet point: water simultaneously a gas, liquid, and solid

Lissauer & de Pater

The formation of **water droplets** and **ice crystals** make up clouds on Earth



Condensation halts at the **tropopause** where the lapse rate turns over



Produces a “**cloud anvil**”

Clouds in the solar system

Giant planets:

NH_3 , H_2S , CH_4

Mars:

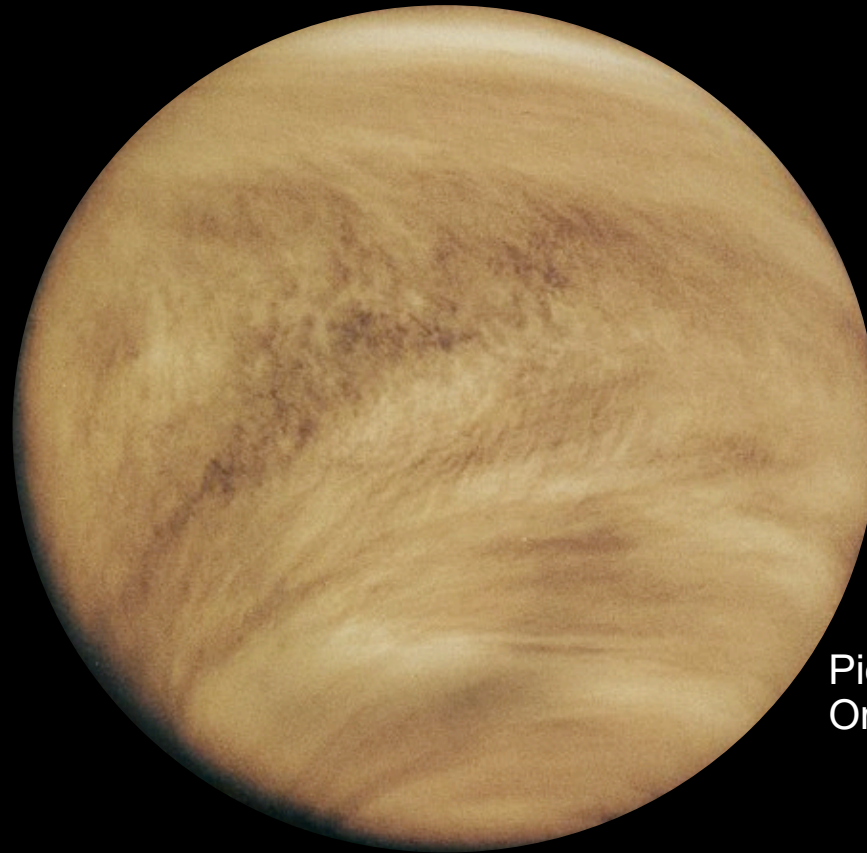
CO_2

Venus:

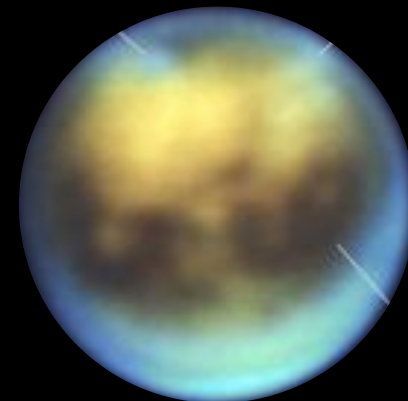
H_2SO_4

Titan (largest of Saturn's moons):

CH_4 , C_2H_6 , + other complex hydrocarbons



Pioneer Venus Orbiter



Keck/NIRC2

Can we detect clouds on **exoplanets**?

Yes, via transmission spectroscopy

Absorption & Emission

Continuous light source

Cloud of gas

Light

CONTINUOUS SPECTRUM

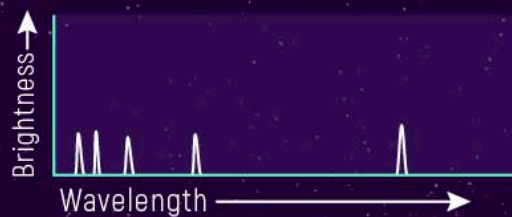
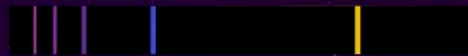
Spectrum that contains **all wavelengths** emitted by a hot, dense, light source



blackbody

EMISSION SPECTRUM

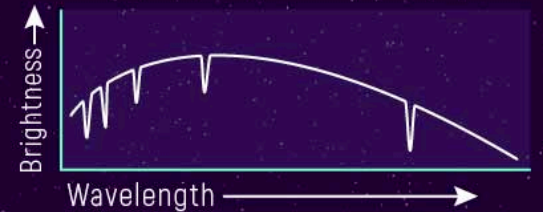
Shows **colored lines** of light emitted by glowing gas



Hot gas

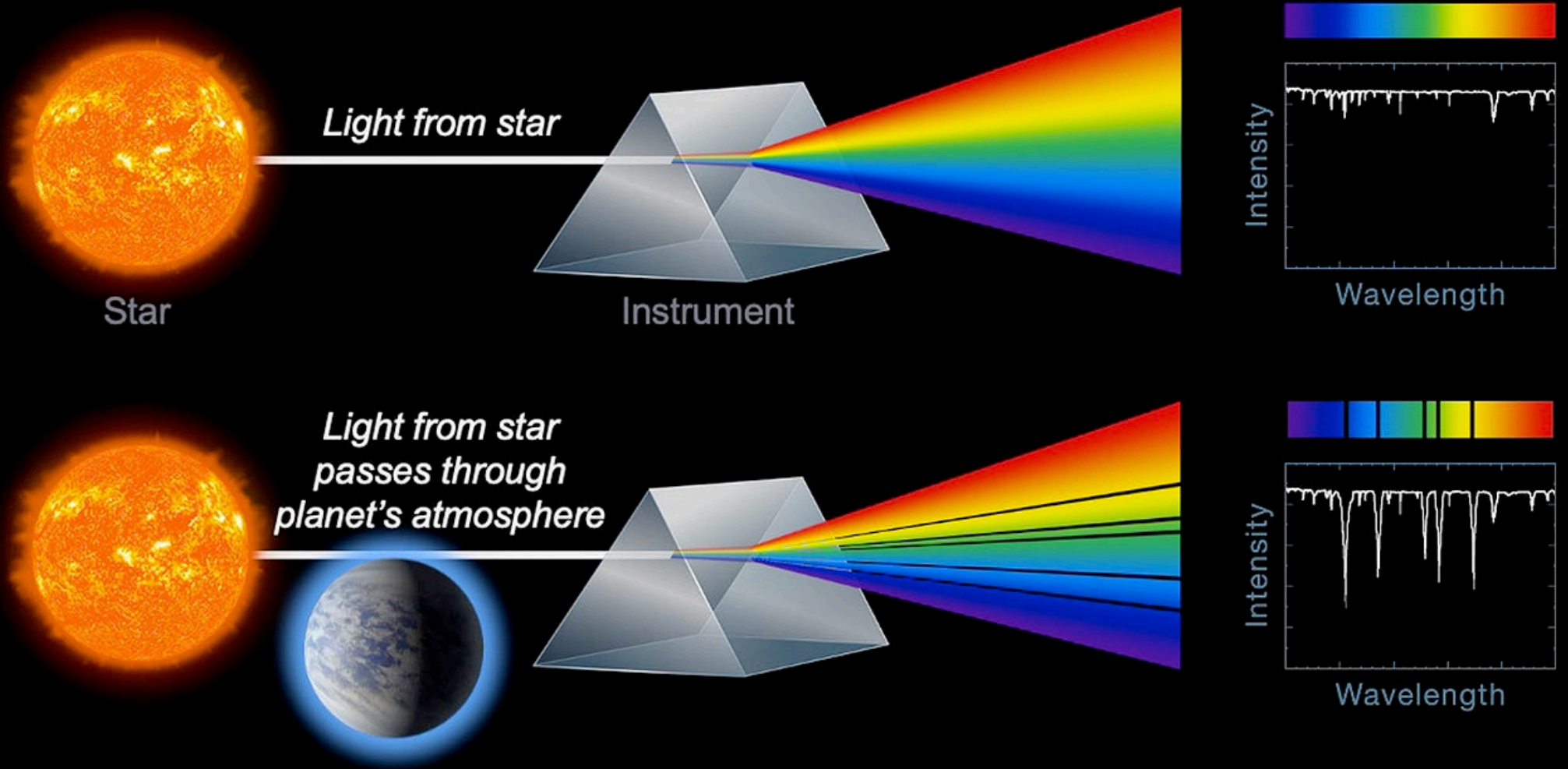
ABSORPTION SPECTRUM

Shows **dark lines or gaps** in light after the light passes through a gas



Cold gas

Transmission Spectroscopy



November 1, 2021

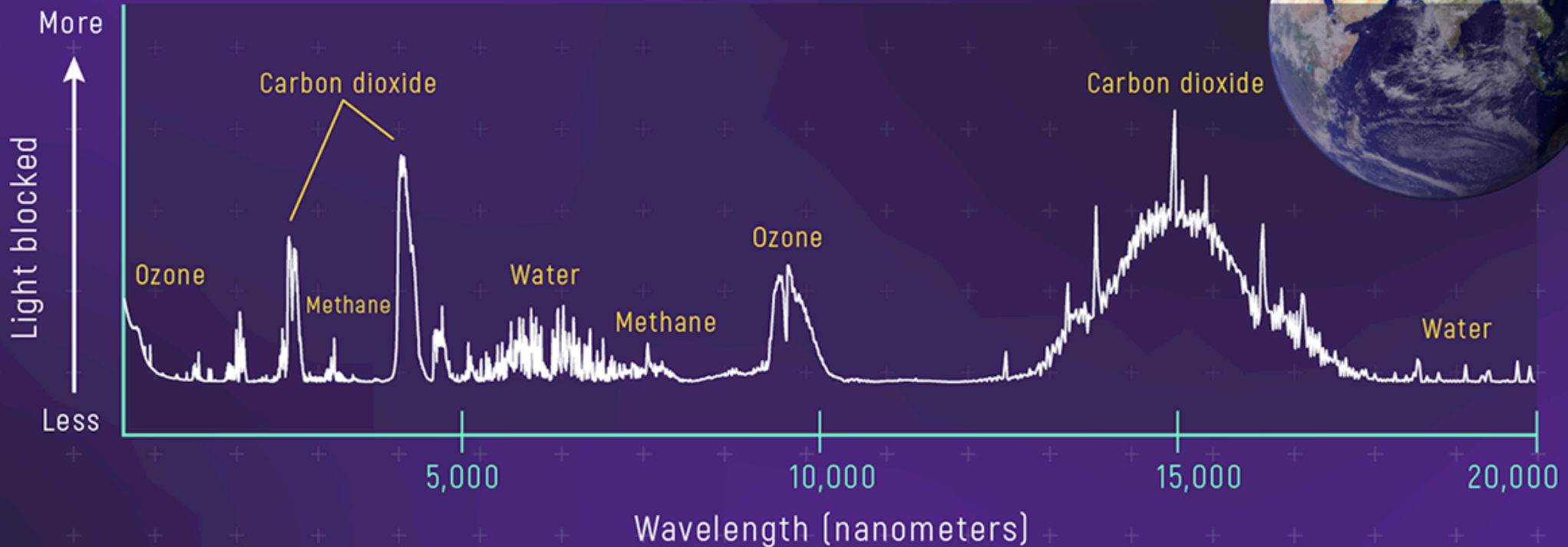
This document has been reviewed and determined not to contain export controlled technical data.

1

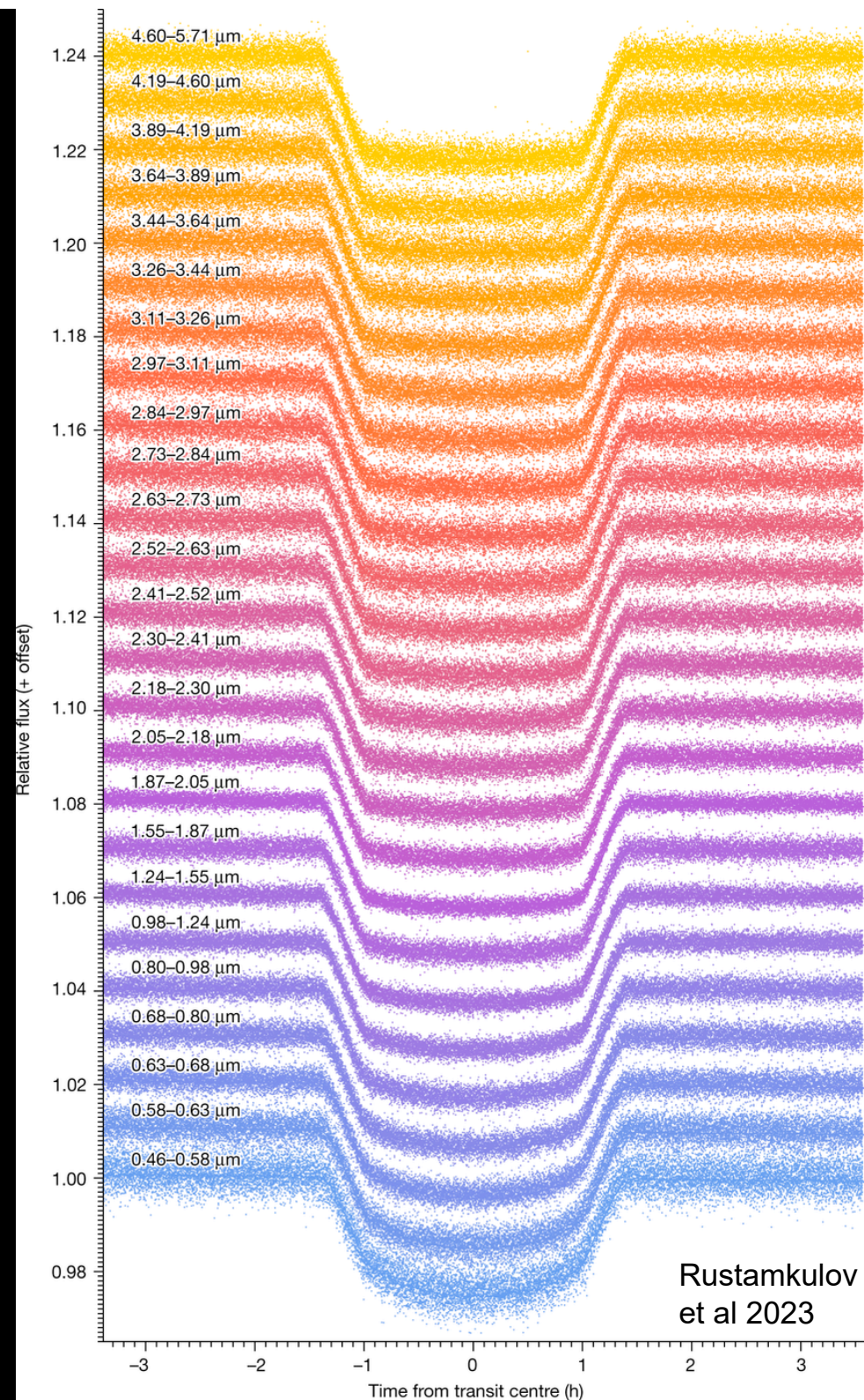
exoplanets.nasa.gov

Transmission Spectroscopy traces atmospheric composition

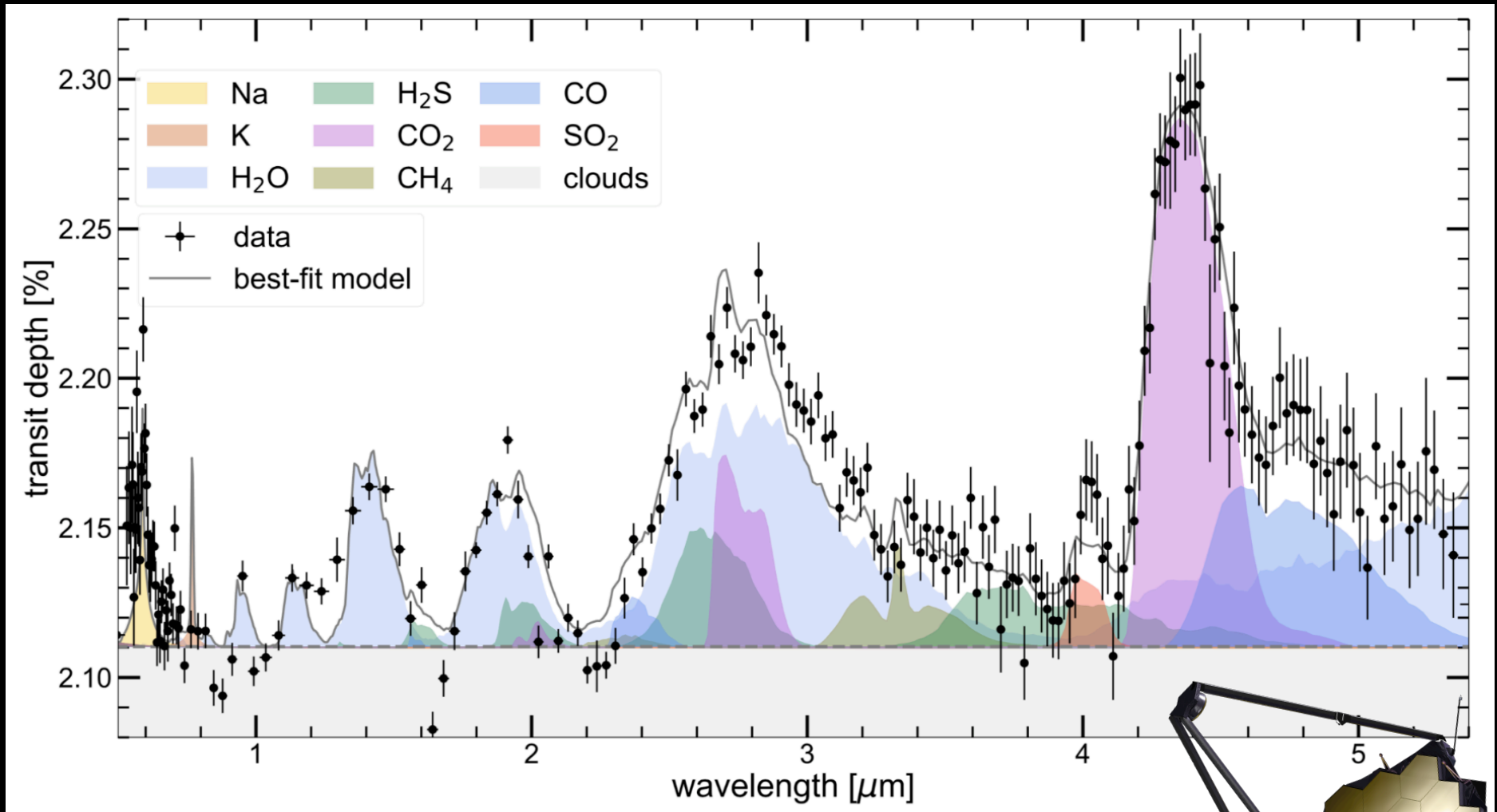
TRANSMISSION SPECTRUM OF AN EARTH-LIKE ATMOSPHERE



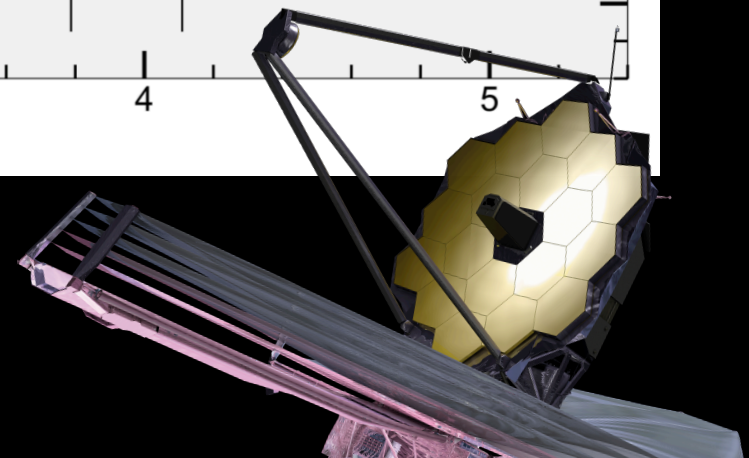
Transmission Spectroscopy = transit depth as a function of wavelength



Transmission Spectrum of the hot Jupiter WASP-39 b



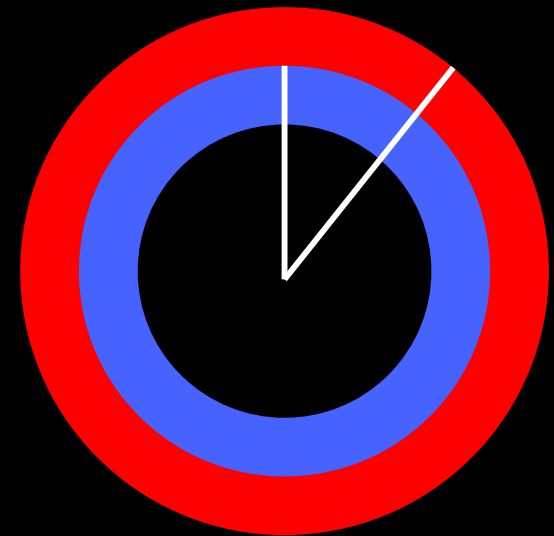
Rustamkulov et al 2023



Absorption depth ΔZ scales with atmospheric scale height H

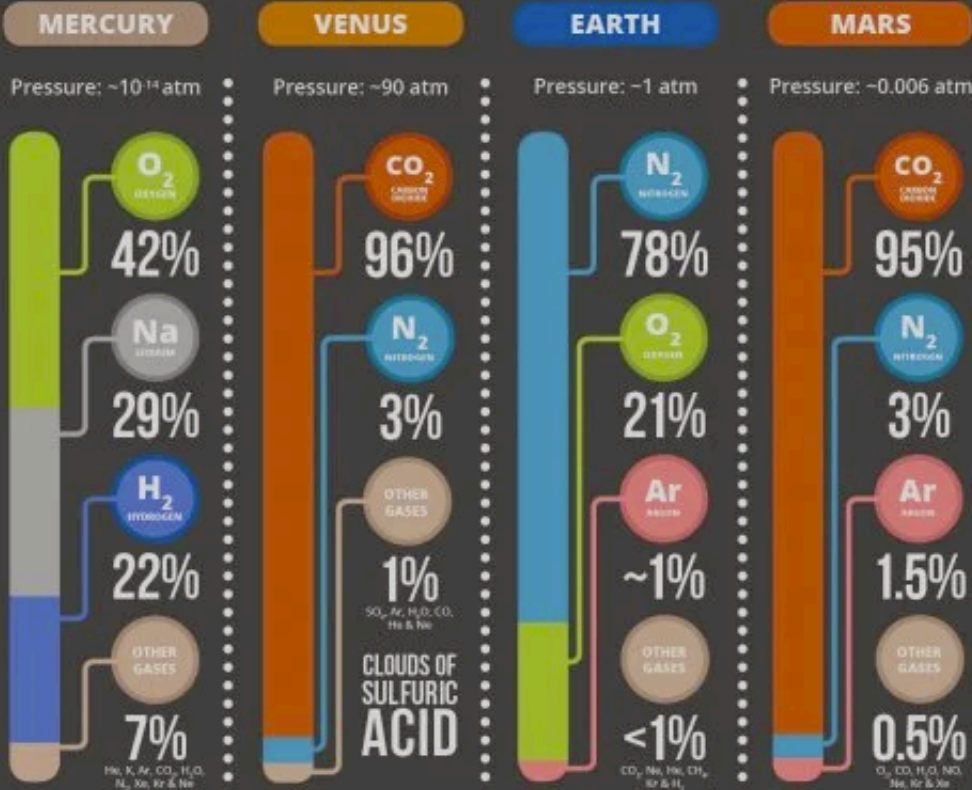
$$H = \frac{k_B T}{\mu m_H g}$$

$$\Delta Z = \frac{\pi (R_p + nH)^2}{\pi R_\star^2} - \frac{\pi R_p^2}{\pi R_\star^2}$$
$$\approx 2 \left(\frac{R_p}{R_\star} \right)^2 \left(\frac{nH}{R_p} \right)$$

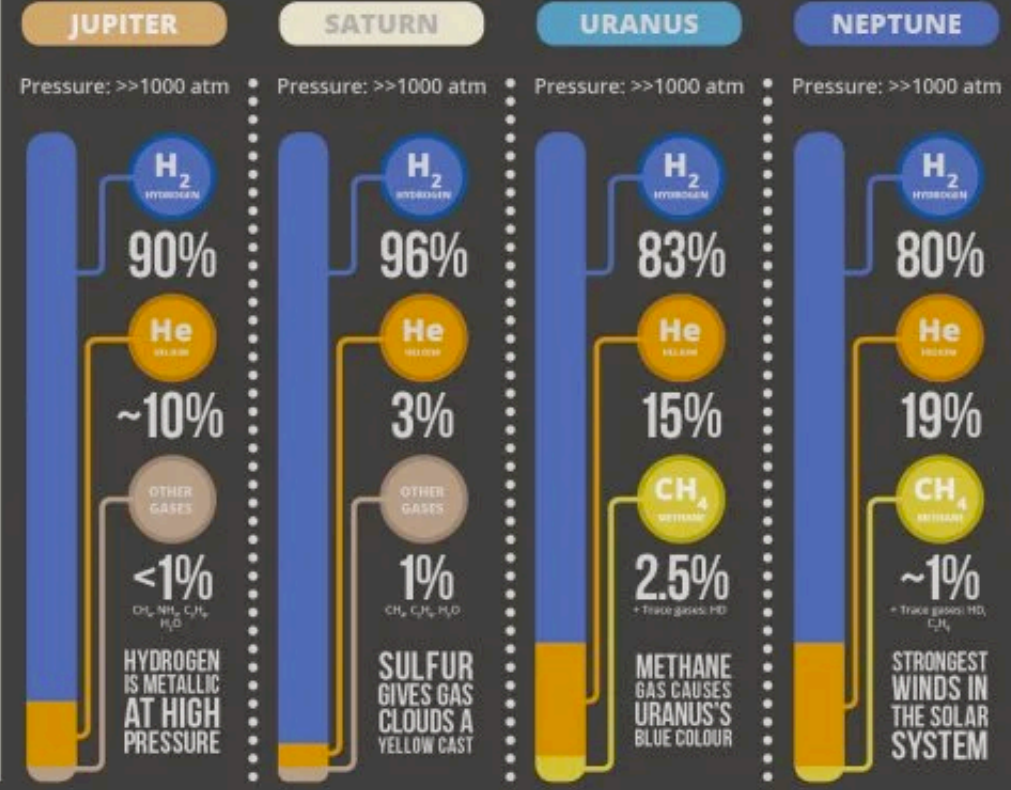


Atmospheric Composition in the solar system

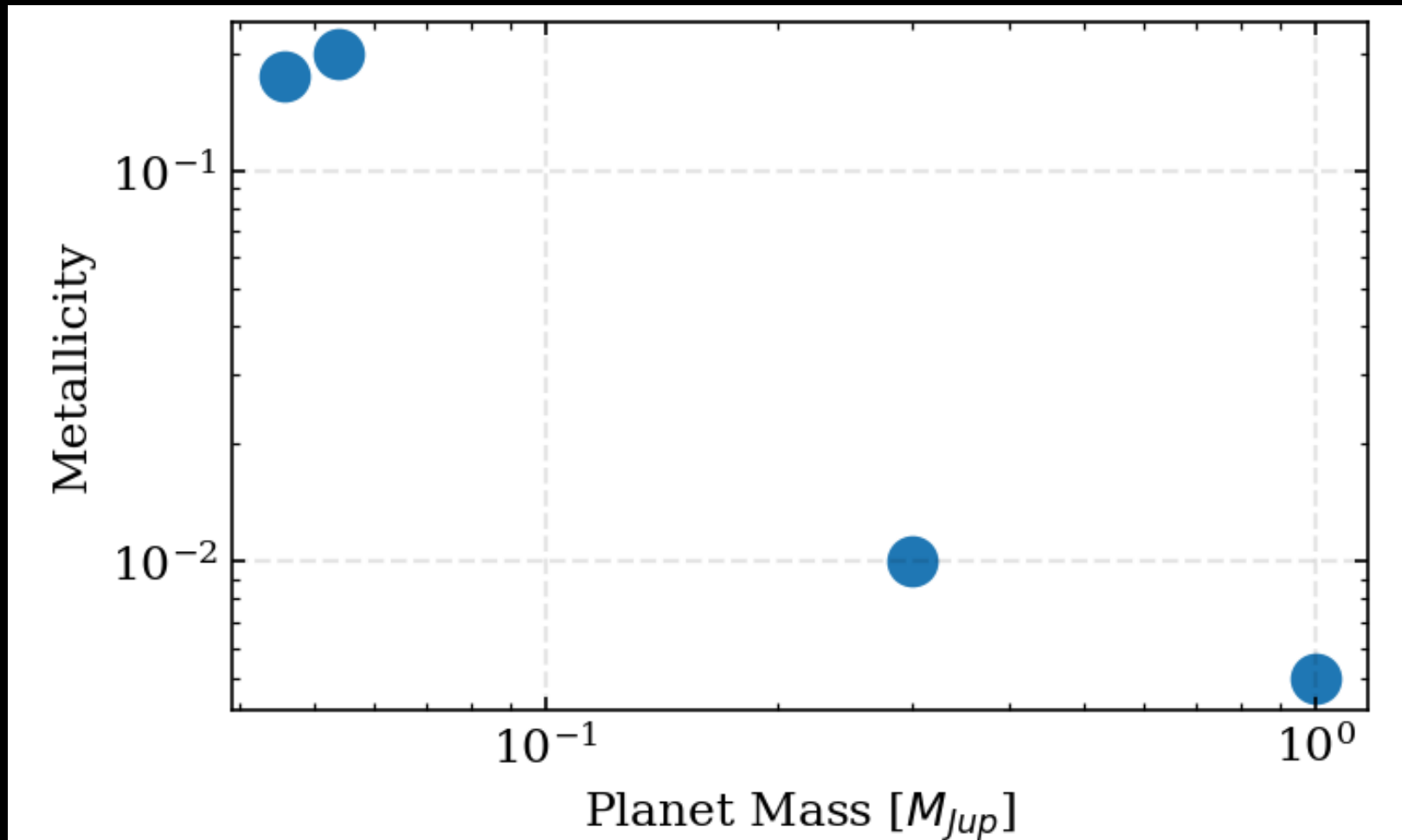
The Terrestrial Planets



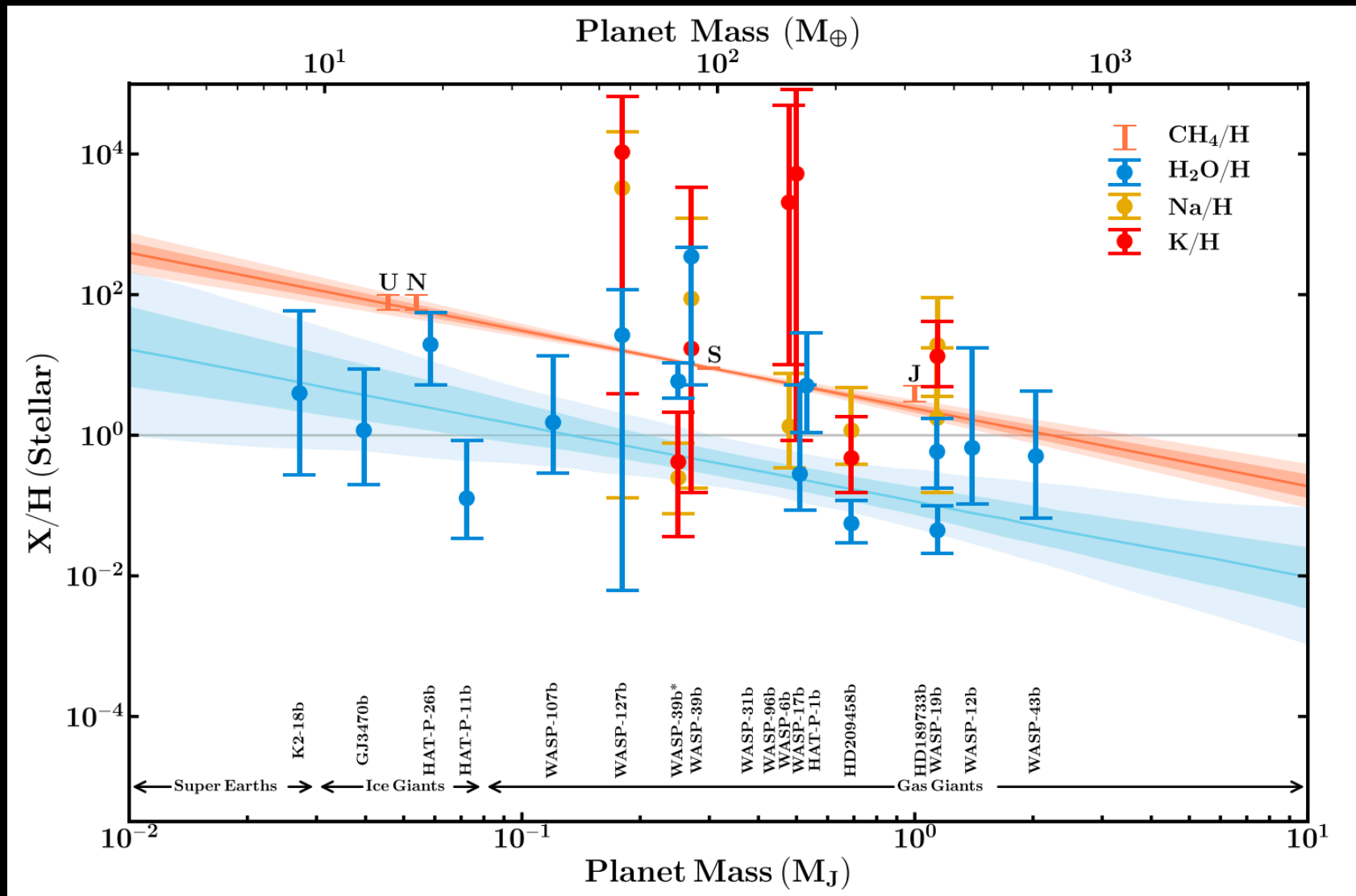
The Gas Giants



The gas giants in the solar system exhibit a **mass-metallicity trend**



Maybe the **mass-metallicity trend** extends to exoplanets...

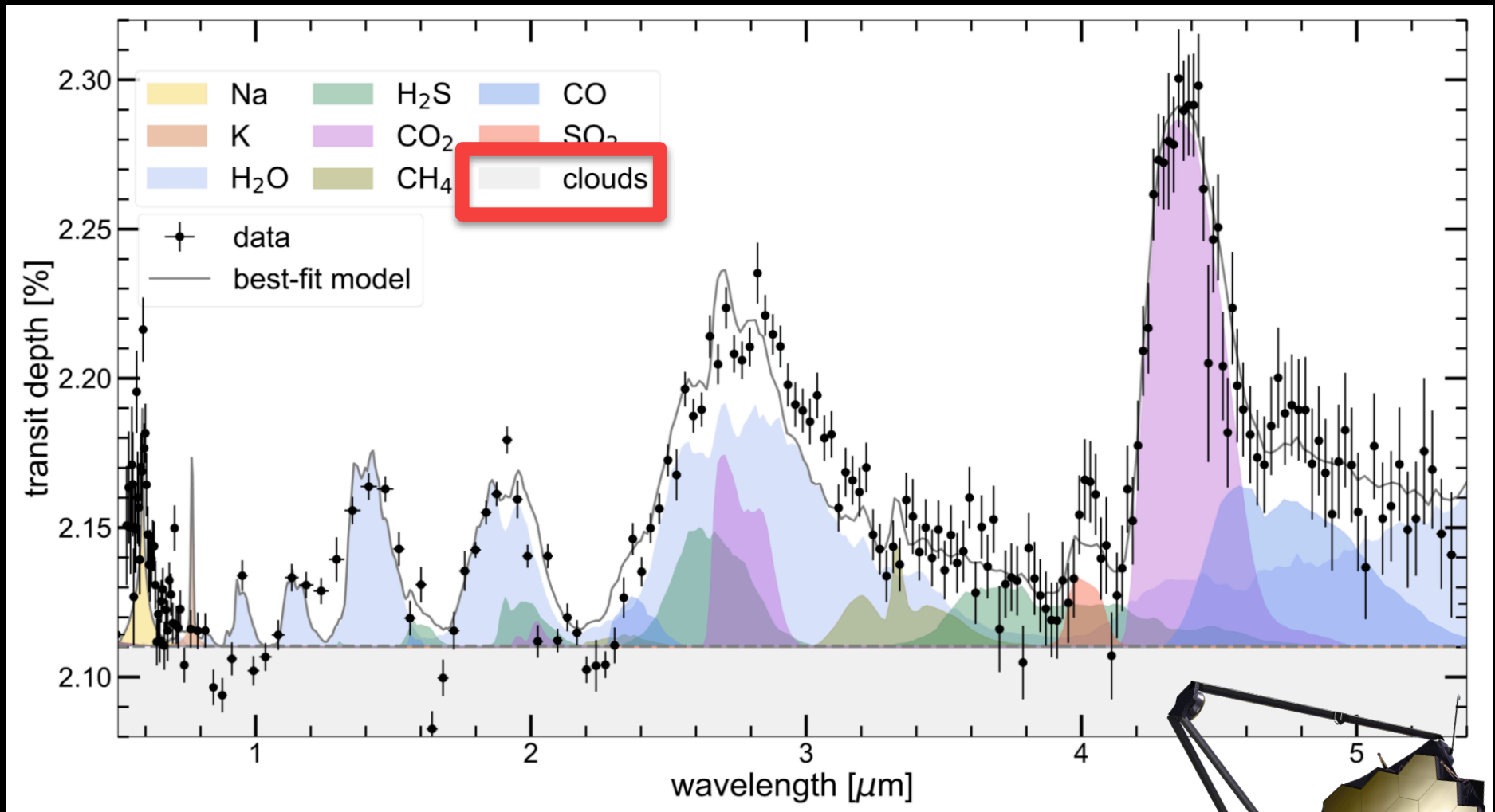


Wellbanks et al 2019

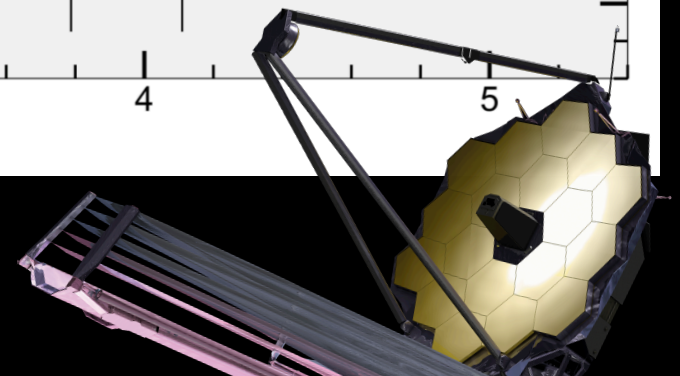
Recall the initial question that started our discussion on transmission spectroscopy...

Can we detect clouds on **exoplanets?**

Clouds **suppress** atmospheric chemical signatures

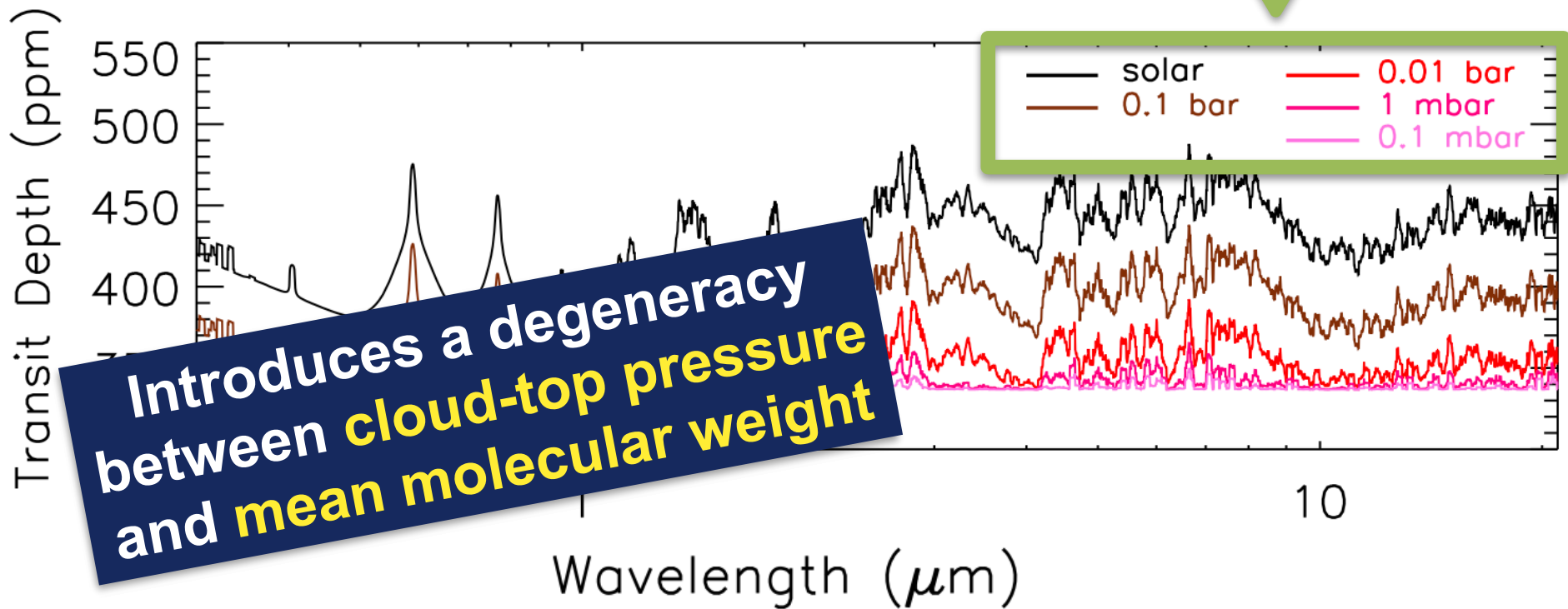


Rustamkulov et al 2023



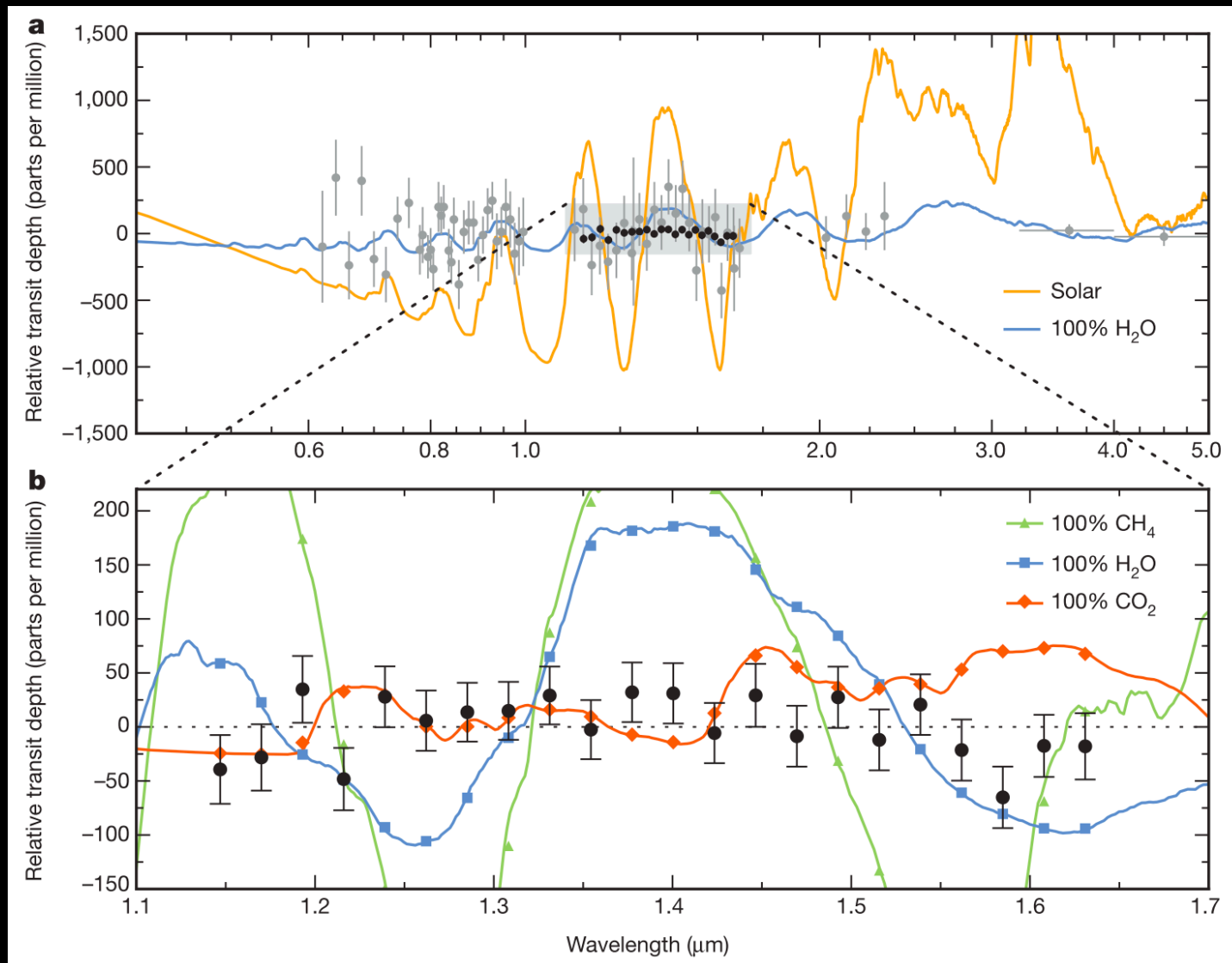
Clouds **suppress** atmospheric chemical signatures

Cloud deck **pressures**



Kempton et al 2016

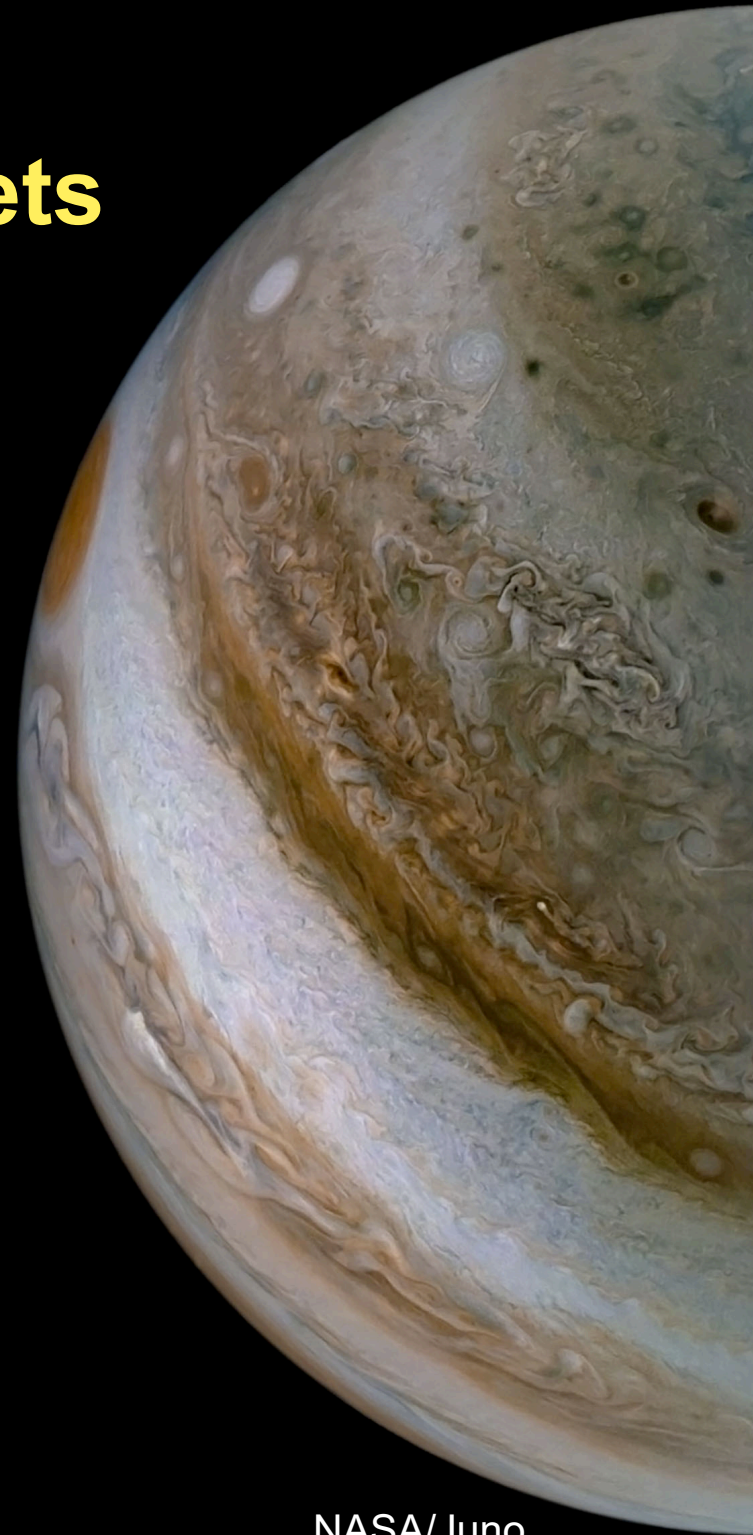
Clouds on the sub-Neptune exoplanet GJ 1214 b



Kreidberg et al 2014

Clouds/winds on **Giant Planets**

- non-water species (e.g. NH_3 , H_2S , CH_4)
- Forms **zones** and **belts**. **Why?**

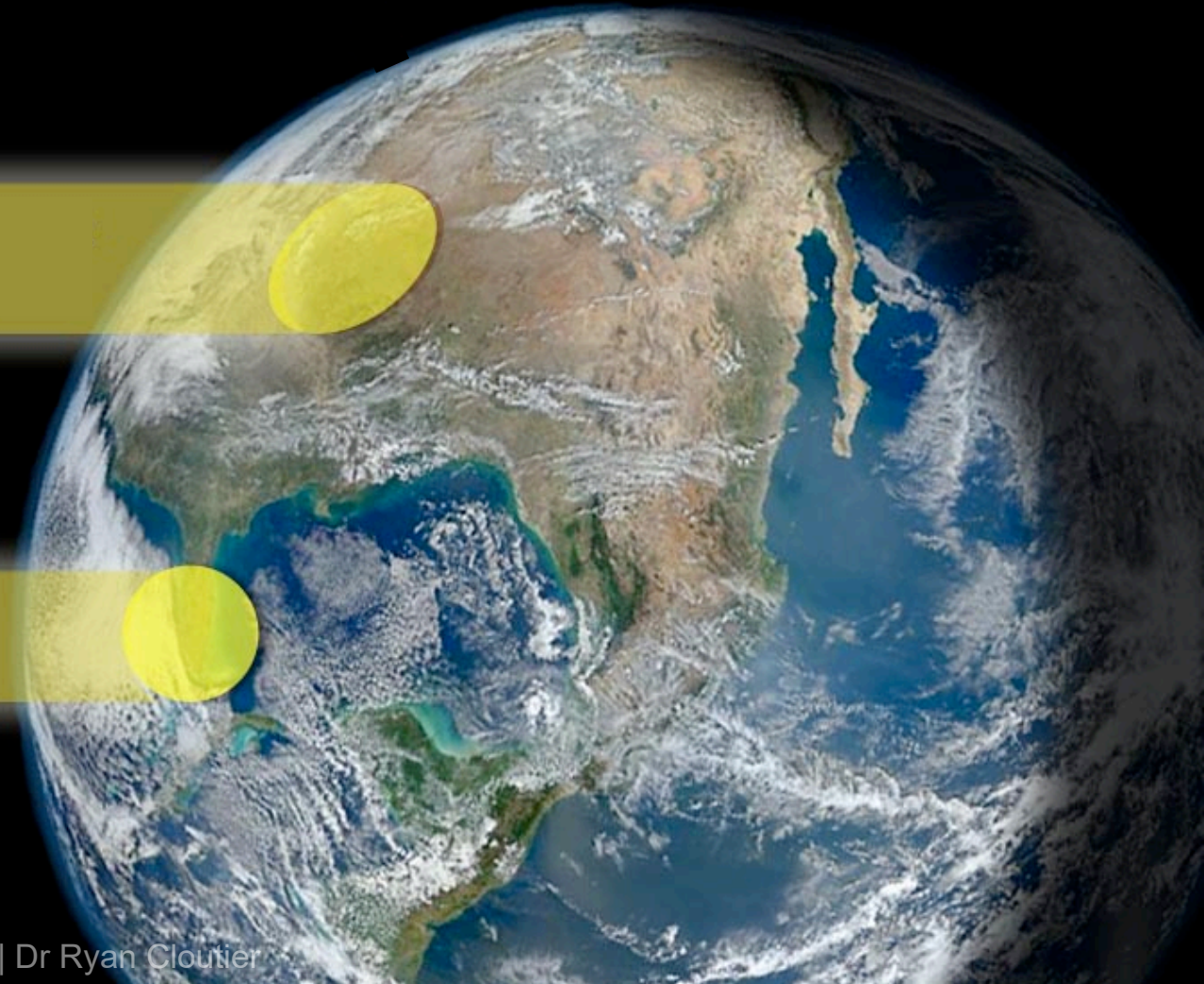


Incident flux on a surface patch depends on the **incident angle**

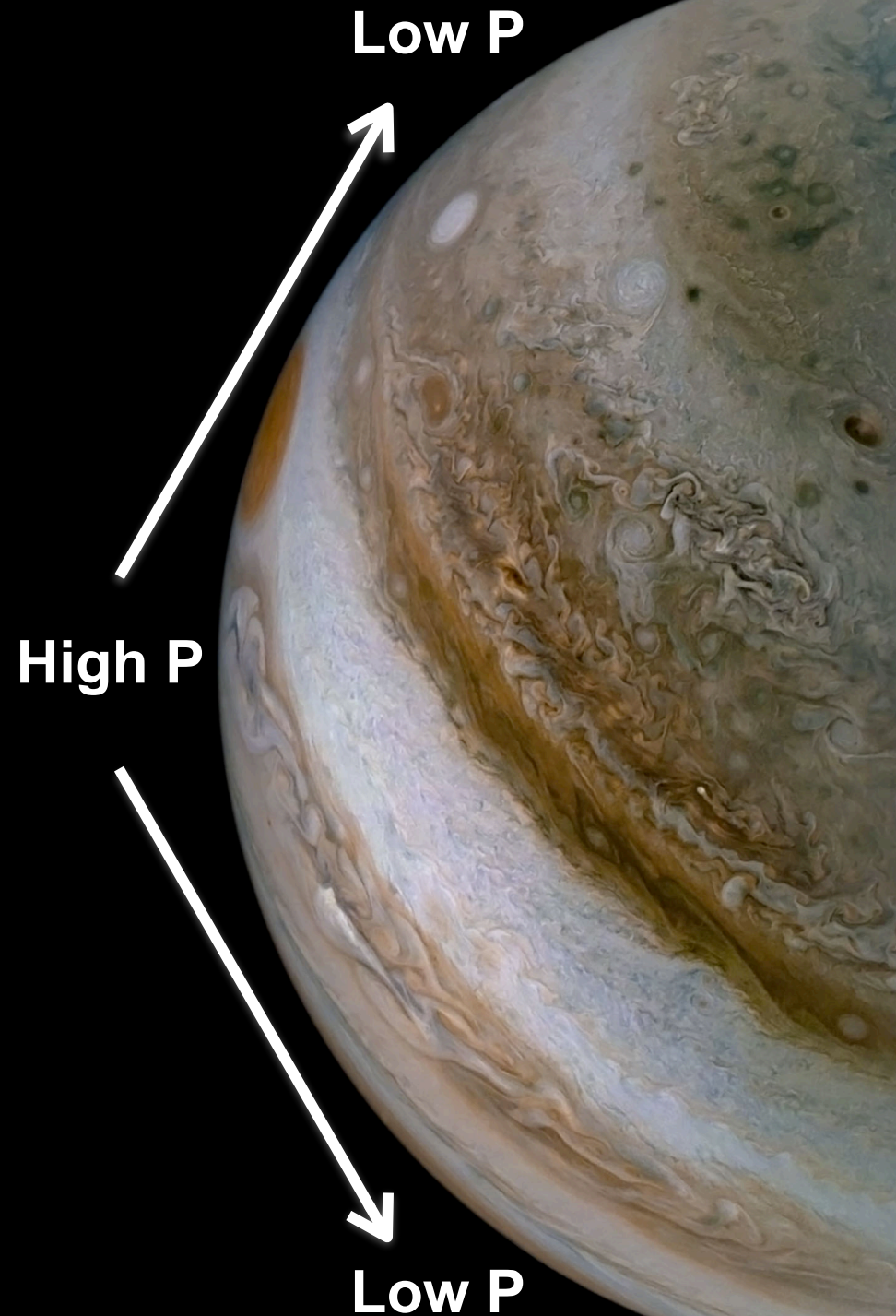
Stronger heating at the equator compared to high latitudes



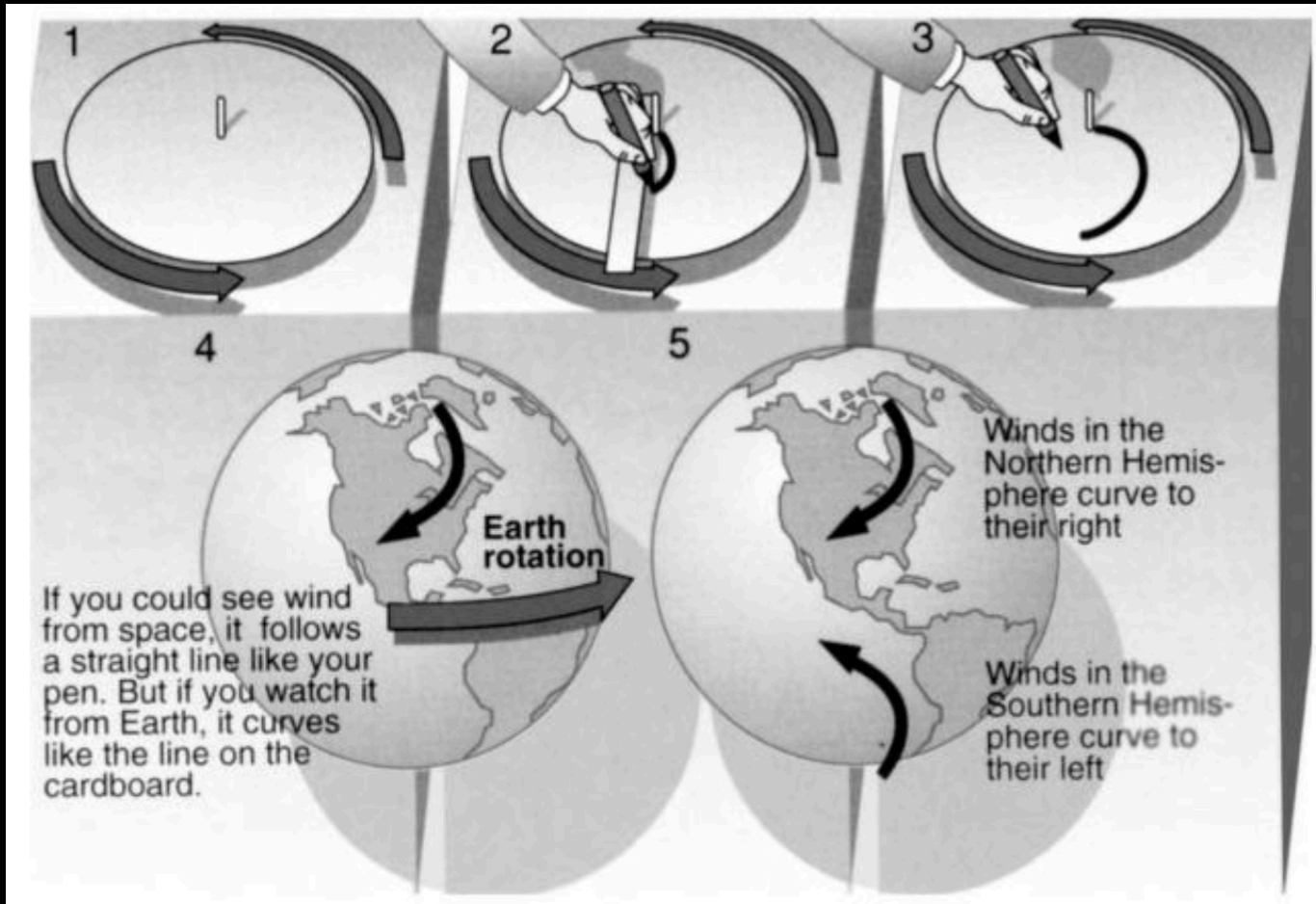
Pressure gradient that drives winds



- Hot equatorial air rises and **flows toward low pressure regions** (i.e. high latitudes)
- Air cools, subsides, and **flows back to the equator**



Coriolis Effect: winds do not follow a straight trajectory

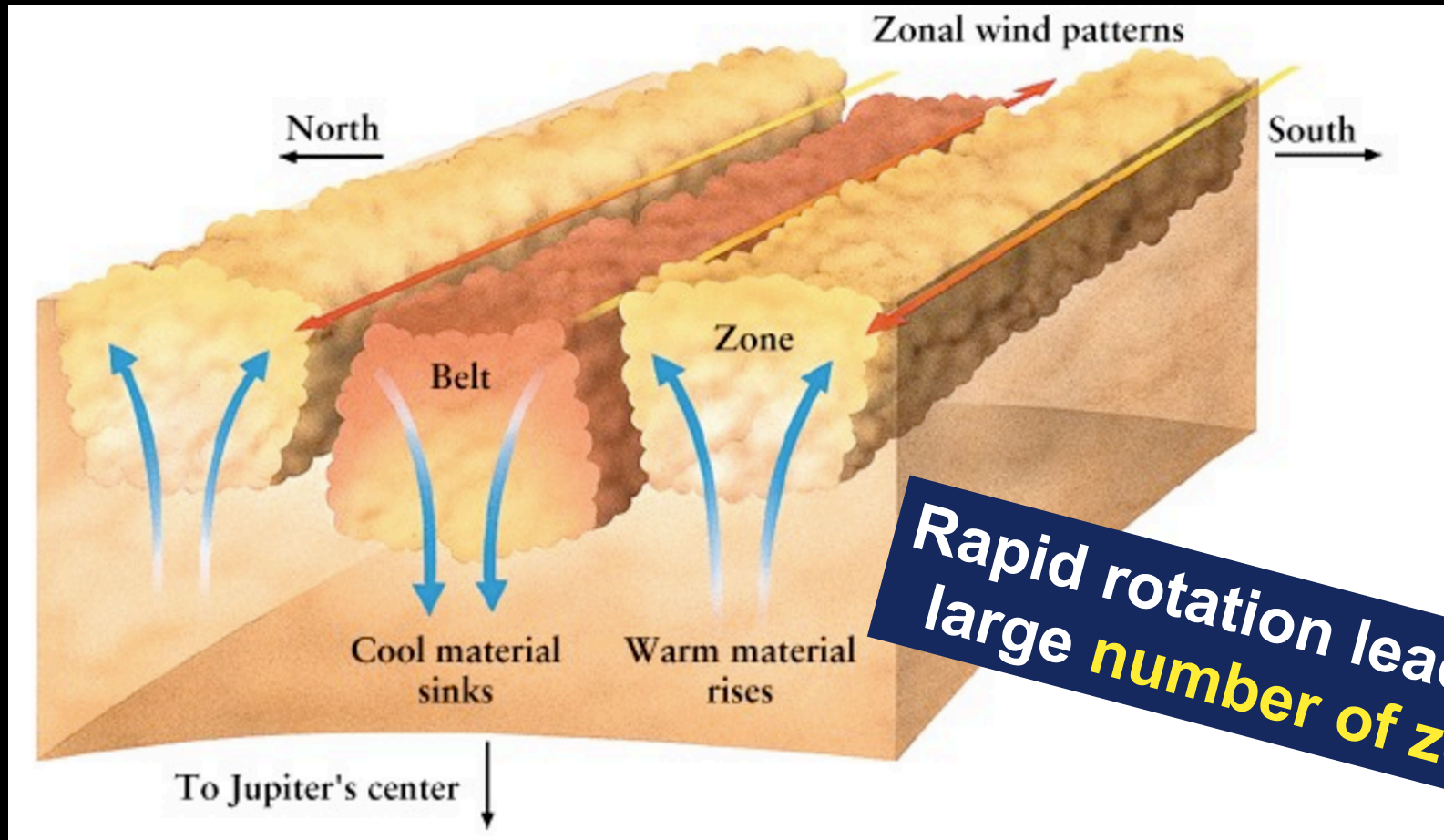


Lissauer & de Pater

On a rotating planet, winds traveling from the poles will be **deflected to the westward** (if the planet's rotation is **prograde**)

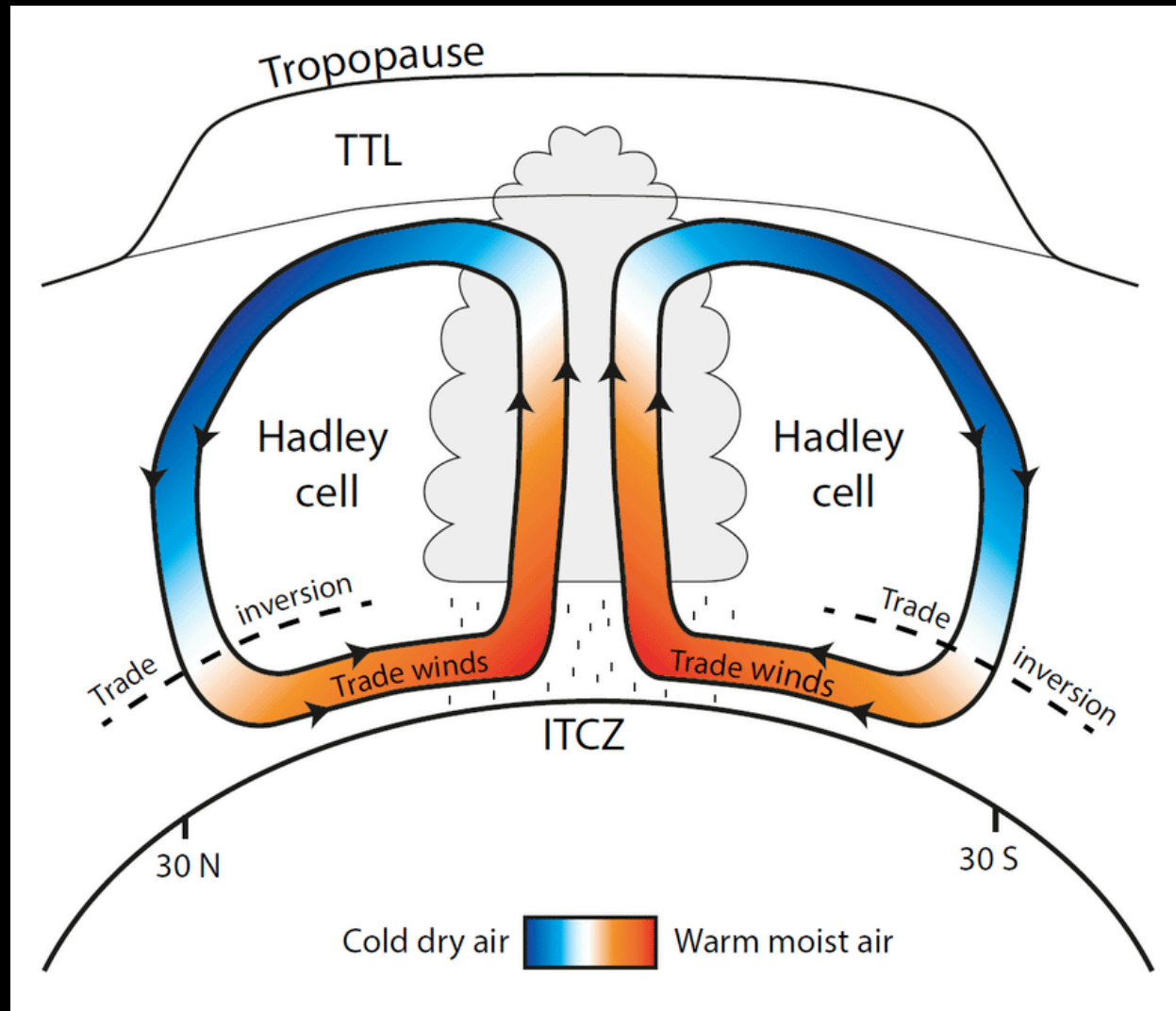
Winds on Giant Planets

Pressure gradient + rapid rotation → zonal winds



Winds on Terrestrial Planets

- Earth ($P_{rot} = 24$ hrs) has **three wind cells** per hemisphere
- Venus ($P_{rot} = 2802$ hrs) has **one large wind cell** per hemisphere

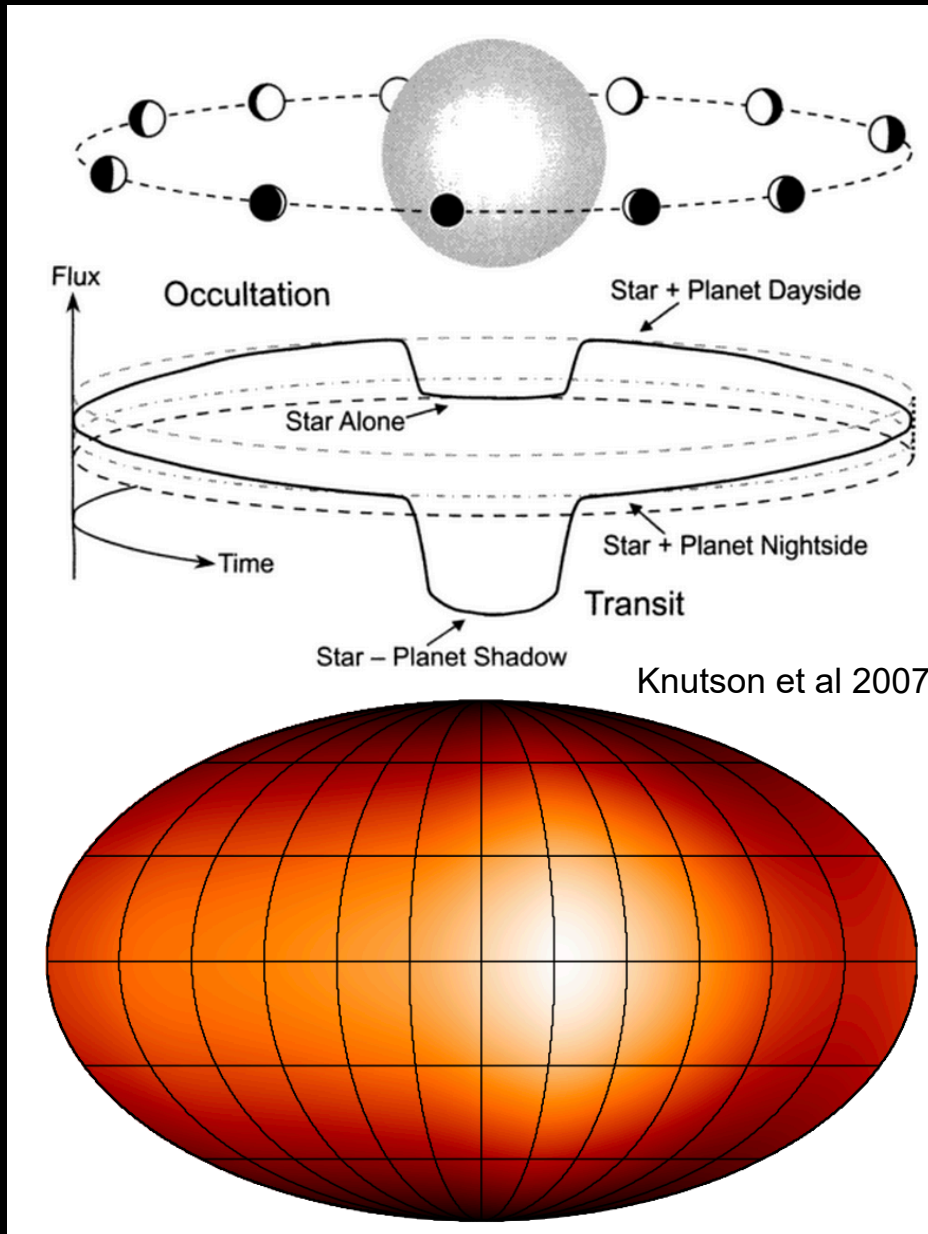


Credit: Alina Fiehn

Winds on **Exoplanets**

For the majority of exoplanets, planetary **rotation rates are not observable**,
but there are a couple of techniques that
work for certain exoplanets

Winds on tidally-locked exoplanets



Recall that tidally-locked planets have **permanent** (i.e. hot) **daysides**

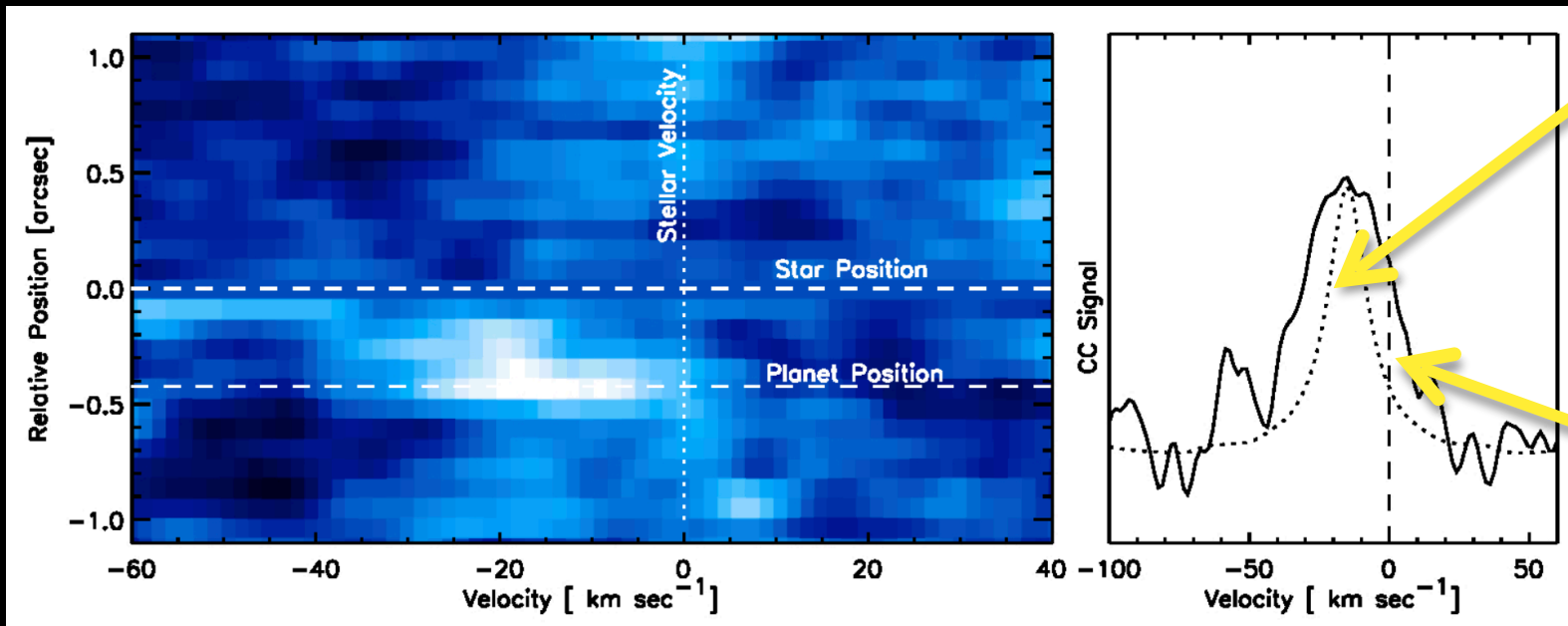
Observations around the secondary eclipse reveal a **hot-spot offset**



Atmospheric circulation

Winds on **directly-imaged exoplanets**

Rotation broadens spectroscopic signals



Signal for a non-rotating planet

Observed signal shows **rotational broadening**

Snellen et al 2014